Reg. No. :

# Question Paper Code : 11333

B.E./B.Tech. DEGREE EXAMINATION, APRIL/MAY 2011

**Fifth Semester** 

**Electronics and Instrumentation Engineering** 

EI 2302 — ANALYTICAL INSTRUMENTS

(Common to Instrumentation and Control Engineering)

(Regulation 2008)

**Time : Three hours** 

Maximum : 100 marks

Answer ALL questions

# PART A — $(10 \times 2 = 20 \text{ marks})$

- 1. State Beer-Lambert's law.
- 2. What are the advantages of double beam spectrophotometer?
- 3. Define Thin Layer Chromatography.
- 4. What is meant by column packing in chromatography?
- 5. Write the principle of a smoke meter.
- 6. Mention the uses of IR analyzer.
- 7. What are called bio-sensors?
- 8. Give the principle of pH measurement.
- 9. What is the advantage of electron spin Resonance Spectroscopy?
- **10.** Write down the principle of Mass Spectrometry.

### **PART B** — $(5 \times 16 = 80 \text{ marks})$

11. (a) With a schematic diagram, explain the Atomic absorption spectroscopy.

Or

- (b) With a neat instrumentation setup, explain the visible spectrophotometer.
- 12. (a) (i) With the necessary diagram, explain the adsorption chromatography techniques. (12)
  - (ii) List the various detectors used for this purpose. (4)

Or

- (b) (i) With a neat schematic diagram, discuss the separation principle of HPLC (High Pressure Liquid Chromatography). (12)
  - (ii) Enumerate the applications of HPLC. (4)
- 13. (a) (i) With a suitable diagram, explain the construction and working of oxygen analyzer. (10)
  - (ii) Discuss the method of analysis based on ionization of gases. (6)

#### Or

- (b) (i) With necessary diagram, explain the working principle of thermal conductivity analyzer.
  - (ii) With neat instrumentation setup, explain any one method to estimate sulphur- di- oxide present in air. (8)
- 14. (a) Discuss, how pH of a solution is measured using glass electrode and reference electrode with necessary diagrams. (16)

#### Or

- (b) Explain the working principle of sodium analyzer and dissolved oxygen analyzer with neat diagram.
- 15. (a) Describe, how various samples are analyzed using NMR spectrometer with neat sketch.

## Or

(b) With neat sketch, explain the construction and working principle of mass spectrometer.

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