



SAMPLE PAPER

FOR CLASS **12th Pass Students**

STREAM: [MEDICAL]

TIME: 2 Hours

FULL MARKS: 480

INSTRUCTIONS

[A] General

- 1. This Question paper contains THREE Parts, A, B and C (Physics, Chemistry, and Biology).
- 2. This Question Paper contains 20 pages including cover page.
- 3. This question paper contains total 120 questions (30 Question in Physics 30 in Chemistry and 60 Question in Biology)
- 4. The Question Paper has blank spaces at the bottom of each page for rough work. No additional sheets will be provided for rough work.
- 5. Blank papers, clip boards, log tables, slide rule, calculators, cellular phones, pagers and electronic gadgets, in any form, are NOT allowed.
- 6. The OMR (Optical Mark Recognition) sheet shall be provided separately.

[B] Answering on the OMR

- 7. In all the parts, each question will have 4 choices out of which only one choice is correct.
- 8. Darken the bubble with Ball Pen (Blue or Black) ONLY.

[C] Filling OMR

- 9. On the OMR sheet, fill all the details properly and completely, otherwise your OMR will not be checked.
- 10. Do not write anything or tamper the barcode in the registration no. box.

[D] Marking Scheme:

11. For each question you will be awarded 4 marks if you darken the bubble corresponding to the correct answer ONLY and zero (0) marks if no bubble is darkened. In all other cases, minus one (–1) mark will be awarded.

Name :	 	 	 	 	
Registration No.:					



SECTION – A: PHYSICS

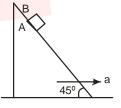
- 1. A ball is thrown vertically upward with a speed v from a height h meter above the ground. The time taken for the ball to hit ground is

- (A) $\frac{v}{g}\sqrt{1-\frac{2gh}{v^2}}$ (B) $\frac{v}{g}\sqrt{1+\frac{2gh}{v^2}}$ (C) $\sqrt{1+\frac{2gh}{v^2}}$ (D) $\frac{v}{g}\left[1+\sqrt{1+\frac{2gh}{v^2}}\right]$
- A particle moves along the positive branch of the curve $y = \frac{x^2}{2}$ where $x = \frac{t^2}{2}$ xand y are measured 2.

in metres and t in second .At t = 2 s, the velocity of the particle is

- (A) $2\hat{i} 4\hat{j} \text{ m/s}$
- (B) $4\hat{i} + 2\hat{j} \, m/s$
- (C) $2\hat{i} + 4\hat{j} \text{ m/s}$
- (D) 4î-2ĵ m/s
- A ballon of weight w is falling vertically downward with a constant acceleration a (<g). The 3. magnitude of the air resistance is
 - (A) w
- (B) $w\left(1+\frac{a}{g}\right)$ (C) $w\left(1-\frac{a}{g}\right)$
- 4. If the coefficient of friction between A and B is μ , the maximum acceleration of the wedge A for which B will remain at rest with respect to the wedge is
 - (A)

(C) $g\left(\frac{1-\mu}{1+\mu}\right)$

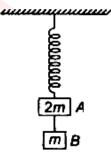


- 5. Two blocks of masses 2m and m are in equilibrium as shown in the figure. Now the string between the blocks is suddenly broken. The accelerations of the blocks A and B respectively at that instant are
 - (A) g and g

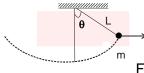
(B) g and $\frac{9}{2}$

(C) $\frac{g}{g}$ and g

(D) $\frac{g}{2}$ and $\frac{g}{2}$



6. An object of mass m is tied to a string of length L and a variable horizontal force is applied on it which starts at zero and gradually increases until the string makes an angle with the vertical. Work done by the force F is



- (A) $mgL(1-sin\theta)$
- (B) mgL
- (C) mgL $(1-\cos\theta)$
- (D) $2 \text{ mgL}(1-\cos\theta)$

A car is circulating on a circular path of radius r. At some instant its velocity is v and rate of 7. increase of speed is a. The resultant acceleration of the car will be

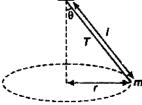


(B) $\sqrt{\frac{V^2}{r} + a}$

(C) $\sqrt{\frac{v^4}{r^2} + a^2}$

(D) $\left(\frac{v^2}{r} + a\right)$

A string of length I fixed at one end carries a mass m at the other end. The string makes $\frac{2}{2}$ rev/s 8. around the axis through the fixed end as shown in the figure, the tension in the string is



(A) 16 ml

(B) 4 ml

(C) 8 ml

(D) 2 ml

The kinetic energy K of a particle moving along a circle of radius R depends on the distance 9. covered s as K = as². The force acting on the particles is

(B) $2as\left(1+\frac{s^2}{R^2}\right)^{1/2}$ (C) $as\left(1+\frac{s^2}{R^2}\right)^{1/2}$

(D) None of these

A simple pendulum is vibrating with an angular amplitude of 90° as shown in 10. the figure. For what value of α , is the acceleration directed?

Vertically upwards (ii) Horizontally

(iii) Vertically downwards

 $0^{\circ}, \cos^{-1}\left(\frac{1}{\sqrt{3}}\right), 90^{\circ}$

(B) $90^{\circ}, \cos^{-1}\left(\frac{1}{\sqrt{3}}\right), 0^{\circ}$

(C) 0° , $\cos^{-1}\sqrt{3}$, 90°

(D) $\cos^{-1}\frac{1}{\sqrt{3}},90^{\circ},0^{\circ}$

An object of mass 3 m splits into three equal fragements. The fragments have velocities VĴ and vî. The velocity of the third fragment is

(A) $v(\hat{J}-v\hat{i})$

(B) $v(\hat{i} - v\hat{J})$ (C) $-v(\hat{i} + \hat{J})$

(D) $\frac{v(\hat{i} + \hat{J})}{\sqrt{2}}$

A body of mass m_1 moving with a velocity 3 ms⁻¹ collides with another body at rest of mass m_2 . After collision the velocities of the bodies are 2 ms⁻¹ and 5 ms⁻¹ respectively along the direction of 12. motion of m_2 . The ratio $\frac{m_1}{m_2}$ is

(A)

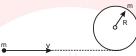
(B) 5

13. The moment of inertia of a semicircular ring of mass M and radius R about an axis which is passing through its centre and at an angle θ with the line joining its ends as shown in figure is



- (a) $\frac{MR^2}{4}$ at $\theta = 0^0$ (b) $\frac{MR^2}{2}$ if $\theta = 0^0$ (c) $\frac{MR^2}{2}$ if $\theta = 45^0$ (d) $\frac{MR^2}{2}$ if $\theta = 90^0$ (A) a, b, c (B) b. c. d (C) a, c. d

- A circular disc of mass m and radius R rests flat on a horizontal frictionless surface. A bullet, also of mass m and moving with a velocity v, strikes that disc and gets embedded in it.

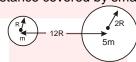


The angular velocity with which the system rotates after the bullet strikes the hoop is

- (C) $\frac{2v}{3R}$
- From a circular disc of radius R and mass 9 M, a small disc of radius $\frac{R}{3}$ is removed from the disc, 15. the moment of inertia of the remaining disc about an axis perpendicular to the plane of the disc and passing O is

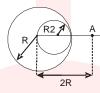


- (B) $\frac{40}{9} MR^2$
- (C) 10 MR²
- (D) $\frac{37}{9}$ MR²
- Two spherical bodies of masses m and 5 m and radii R and 2R respectively are released in free 16. space with initial separation between their centers equal to 12 R. If they attract each other due to gravitational force only then the distance covered by smaller sphere just before collision will be



- (A) 5 R
- (B) 7.5 R
- (D) R
- Two particles of mass m and M are initially at rest at infinite distance. Find their relative velocity of approach due to gravitational attraction when d is separation at any instant

18. A solid sphere of uniform density and radius R applies a gravitational force of attraction equal to F₁ on a particle placed at a distance 2R from the centre of the sphere. A spherical cavity of radius R/2 is now made in the sphere as shown in the figure. The sphere with cavity now applies a gravitational force F₂ on the same particle. The ratio F/2 is



- (A) $\frac{5}{9}$
- (B) $\frac{7}{8}$
- (C) $\frac{3}{4}$
- (D) $\frac{7}{9}$
- 19. Gravitational field at the centre of a semicircle formed by a thin wire AB of mass M and length I is

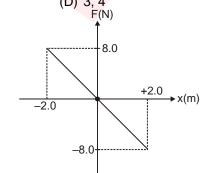


(A) $\frac{GM}{\ell^2}$ along x – axis

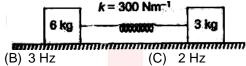
(B) $\frac{\text{GM}}{\pi \ell^2}$ along y – axis

(C) $2\pi \frac{GM}{\ell^2}$ along x – axis

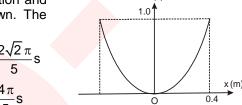
- (D) $\frac{2\pi GM}{\ell^2}$ along y axis
- 20. The displacement equation of a particle is $x = 3 \sin 2t + 4 \cos 2t$. The amplitude and maximum velocity will be respectively
 - (A) 5, 10
- (B) 3, 2
- (C) 4, 2
- 21. A body of mass 0.1 kg executes simple harmonic motion (SHM) about x = 0 under the influence of a force shown in figure. The period of the SHM is
 - (A) 0.99 s
 - (B) 0.52 s
 - (C) 0.25 s
 - (D) 0.31 s



22. Two point masses of 3.0 kg and 6.0 kg are attached to opposite ends of horizontal spring whose spring constant is 300 Nm⁻¹ as shown in the figure. The natural vibration frequency of the system is approximately



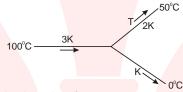
(A) 4 Hz A particle of mass 2 kg moves in simple harmonic motion and 23. its potential energy U varies with position x as shown. The period of oscillation of the particle is



(D) 1 Hz

- (B) $\frac{2\sqrt{2}\pi}{5}$ s
- (C) $\frac{\sqrt{2}\pi}{5}$ s

- (D) $\frac{4\pi}{5}$ s
- A cylinder of radius r and of thermal conductivity K₁ is surrounded by a cylindrical shell of inner 24. radius r and outer radius 2r made of a material of thermal conductivity K2. The effective thermal conductivity of the system is
- (A) $\frac{1}{3}(K_1 + 2K_2)$ (B) $\frac{1}{2}(2K_1 + 3K_2)$ (C) $\frac{1}{3}(3K_2 + 2K_1)$ (D) $\frac{1}{4}(K_1 + 3K_2)$
- Three rods of same dimensions have thermal conductivities 3 K, 2 K, and K,. They are arranged 25. as shown below



What will be temperature T of the junction?

- (A) $\frac{200}{3}$ °C
- (B) $\frac{100}{3}$ °C
- (C) 75°C
- The equation of a progressive wave is $y = 8 \sin \left[\pi \left(\frac{t}{10} \frac{x}{4} \right) + \frac{\pi}{3} \right]$. The wavelength of the wave is
 - (A) 8 m

- (D) 10 m
- If two waves represented by $Y_1 = 4 \sin \omega t$ and $y_2 = 3 \sin \left(\omega t + \frac{\pi}{3} \right)$ interfere at a point, the amplitude 27.

of the resulting wave will be about

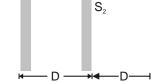
- (A) 7
- (B) 6.08
- (C) 5
- (D) 3.5
- Direction of the first secondary maximum in the Fraunhofer diffraction pattern at a single slit is 28. given by (a is the width of the slit)
 - (A) $a \sin \theta = \frac{\lambda}{2}$
- (B) $a\cos\theta = \frac{3\lambda}{2}$
- (C) $a\sin\theta = \lambda$ (D) $a\sin\theta = \frac{3\lambda}{2}$

- 29. In Young's double slit experiment, white light is used. The separation between the slits is b. The screen is at a distance d(d>>b) from the slits. Some wavelengths are missing exactly in front of one slite. These wavelengths are
 - (A) $\lambda = \frac{b^2}{2d}$
- (B) $\lambda = \frac{2b^2}{d}$
- (C) $\lambda = \frac{b^2}{3d}$
- (D) $\lambda = \frac{2b^2}{3d}$
- 30. Two ideal slits S_1 and S_2 are at a distance d apart and illuminated by light of wavelength λ passing through an ideal source slit S placed on the line through S_2 as shown. The distance between the planes of slits and the sources slit. is D. A screen is held at a distance D from the plane of the slits. The minimum value of d for which there is darkness at O is



(A) $\sqrt{\frac{3\lambda D}{2}}$

- (B) $\sqrt{\lambda D}$
- (D) √3λD



SECTION - B: CHEMISTRY

- 31. The equation which is balanced and represents the incorrect product(s) is :
 - (A) $\text{Li}_2\text{O} + 2\text{KCI} \rightarrow 2\text{LiCI} + \text{K}_2\text{O}$
 - (B) $\left[\text{CoCI} \left(\text{NH}_3 \right)_5 \right]^+ + 5\text{H}^+ \rightarrow \text{Co}^{2+} + 5\text{NH}_4^+ + \text{CI}^-$
 - (C) $\left[Mg(H_2O)_6 \right]^{2+} + \left(EDTA \right)^{4-} \xrightarrow{excess NaOH} \left[Mg(EDTA) \right]^{-2} + 6H_2O$
 - (D) $CuSO_4 + 4KCN \rightarrow K_3 \left[Cu(CN)_4 \right] + K_2SO_4$
- 32. The colour of light absorbed by an aqueous solution of CuSO₄ is
 - (A) Orange-red
- (B) Blue-green
- (C) Yellow
- (D) Violet

33. Among the following complexes (K–P)

 $K_3[Fe(CN_6)](K), [Co(NH_3)_6]CI_3(L), Na_3[Co(ox)_3](M), [Ni(H_2O)_6]CI_2(N)$

 $K_{2}[Pt(CN_{4})](O), [Zn(H_{2}O)_{6}](NO_{3})_{2}(P)$

the diamagnetic complexes are

- (A) K, L, M, N
- (B) K, M, O, P
- (C) L, M, O, P
- (D) L, M, N, O
- 34. Extraction of metal from the ore cassiterite does not involves
 - (A) carbon reduction of an oxide ore
- (B) self-reaction of a sulphide ore
- (C) removal of copper impurity
- (D) removal of iron impurity
- 35. Extraction of copper from copper pyrite (CuFeS₂) does not involves
 - (A) crushing followed by concentration of the ore by froth-flotation
 - (B) removal of iron as slag
 - (C) self-reduction step to produce 'blister copper' following evolution of SO₂.
 - (D) refining of 'blister copper' by carbon reduction
- 36. Which of the following reactions produces N₂O is
 - (i) $Zn+dil.HNO_3 \longrightarrow$

(ii) $NH_4NO_3 \xrightarrow{\Delta}$

(iii) $SnCl_2 + HNO_3 \longrightarrow$

(iv) NO + SO₂ + H₂O \longrightarrow

(v) $NO + H_2S \longrightarrow$

(vi) $NH_2OH \cdot HCI + NaNO_2 \longrightarrow$

(A) only (i) & (ii)

(B) only (i),(ii) & (iii)

(C) only (i),(ii),(iv) & (v)

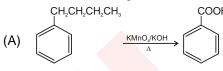
(D) (i), (ii), (iii), (iv), (v) & (vi)

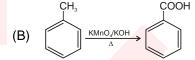
- 37. The correct order of electron given enthalpy is
 - (A) Br > I > C > Si
- (B) Br > I > Si > C
- (C) I > Br > C > Si
- (D) I > Br > Si > C

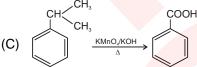
- 38. Total number of Fe atoms in **Prussian's** blue is
 - (A) 4
- (B) 7

- (C) 10
- (D) 6

39. Which of the following reactions does not occur?

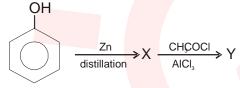






(D)
$$\xrightarrow{\text{KMnO}/\text{KOH}}$$
 $\xrightarrow{\text{COOH}}$

40. Identify the final product of the reaction sequence.



- (A) Benzophenone
- (B) Acetophenone
- (C) Diphenyl
- (D) Methyl salicylate

- 41. In the reaction \bigcirc OCH₃ $\xrightarrow{\text{HBr}}$ the products are:
 - (A) Br OCH₃ and H₂

(B) Br and CH₃Br

(C) Br and CH₃OH

- (D) OH and CH₃Br
- 42. Which of the following sequence of reactions (reagents) can be used for conversion of $C_6H_5 CH_2 CH_3$ into $C_6H_5CH = CH_2$?
 - (A) SOCI₂; H₂O

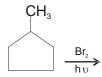
(B) SO₂Cl₂; alc. KOH

(C) Cl₂/hv; H₂O

(D) SOCI₂; alc. KOH

- 43. Which one is correct?
 - (A) Freon-14 is CF₄, Freon-13 is CF₃Cl, Freon-12 is CF₂Cl₂ and Freon-11 is CFCl₃
 - (B) Freons are chlorofluorocarbons
 - (C) Freons are used as refrigerants
 - (D) All the above

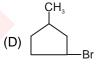
44. In the following reaction, the major product is:









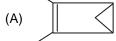


45. In the following reaction, the product 'R' is:

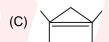
$$CaC_2 \xrightarrow{H_2O} P \xrightarrow{\text{Hot iron}} Q \xrightarrow{\text{CH}_3CI} R$$

- (A) Benzene
- (B) Ethylbenzene
- (C) Toluene
- (D) N-propylbenzene

46. hence X is: [X]



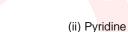




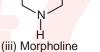


47. Select the basic strength order of following molecules?











(iv) Pyrrole

- (A) (IV) > (I) > (III) > (IV)
- (C) (II) > (I) > (III) > (IV)

(B) (III) > (I) > (IV) > (II) $(D) \quad (I) > (III) > (II) > (IV)$

- 48. Resonance effect involves:
 - (A) Delocalization of π -electrons along a conjugated system
 - (B) Delocalization of lone pair along a conjugated system
 - (C) Delocalization of negative charge along a conjugated system
 - (D) All are correct
- $CH_3 CH_2 CH_2 CH_2 CH_2$ 49.

Inductive effect of chlorine is zero on:

- (A) C_1
- (B) C₂
- (C) C_3
- (D) C₄

50.	Which of the following	g is having maximum	atoms							
	(A) 9.8g H ₂ SO ₄		(B)	342 g Suci	rose					
	(C) 180 g Glucose		(D)	6.72 Litres	of Butane gas at NTP					
51.	Which of the following	g is not a state functi	on ?							
	(A) ΔS	(B) ΔG	(C)	ΔH	(D) Q					
52.	What is $[H^{\dagger}]$ in mol/L CH ₃ COOH = 1.8 × 10		0. <mark>20 M</mark> in C	H₃COONa a	nd 0.10 M in CH_3COOH ? K_a for					
	(A) 3.5×10 ⁻⁴	(B) 1.1×10 ⁻⁵	(C)	1.8×10 ⁻⁵	(D) 9.0×10 ⁻⁶					
53.					<mark>If 5.00</mark> g Na₂SO₄ is dissolved in he van't Hoff factor for Na₂SO₄.					
	(A) 2.05	(B) 2.63	(C)	3.11	(D) 0.381					
54.	Number of neutrons in be	n a <mark>parent nucleus X</mark>	(which give:	s ₇ N ¹⁴ after t	wo successive β-emission would					
	(A) 6	(B) 7	(C)	8	(D) 9					
55.	K _{Sp} of CuS, Ag₂S ar their solubility in wate		0 ⁻⁴⁴ and 10	⁻⁵⁴ respectiv	vely. Select the correct order for					
	(A) $Ag_2S > HgS > C$	uS	(B)	HgS > CuS	$S > Ag_2S$					
	(C) HgS > Ag ₂ S > C	uS	(D)	Ag ₂ S > Cu	S > HgS					
56.	The standard oxidati respectively. The star				e E = - 0.34 V and - 0.16 V					
	(A) 0.18V	(B) 0.52V	(C)	0.82V	(D) 0.49V					
57.	Which of the following has smallest number of molecules?									
	(A) 0.1 moles of CO	₂ gas	(B)	11.2 L of C	CO₂ gas at NTP					
	(C) 22.2 g of CO ₂ ga	as	(D)	$22.4 \times 10^3 \text{r}$	nL of CO₂ gas at NTP					
58.	The average atomic vas compared to the o			The abunda	nce of lighter isotope of mass 50					
	(A) 15%	(B) 45 <mark>%</mark>	(C)	50%	(D) 85%					
59.	8g of a radioactive s substance is:	ubstance is reduced	d to 0.5g aft	ter one hou	r. The half-life of the radioactive					
	(A) 15 min	(B) 30 min	(C)	45 min	(D) 10 min					
60.	For a reaction ΔH be spontaneous ?	= $+3$ kJ and Δ S	= + 10 J/	K at which	temperature this reaction will					
	(A) 210 K	(B) 200 K	(C)	272 K	(D) 272 K					



SECTION – C: BIOLOGY

- 61. Harbarium is one of the important tools that were used for identification of plants. Which one of the following is correct regarding it?
 (A) It provides information about the local flora and fauna of that region.
 (B) The information provided by them are useful in locating wild varieties and relatives of economically important plants
 (C) The new material added to the collection of herbarium is known as acquisition.
 (D) It provides living plant material for systematic work
- (A) Utricularia
 (B) Rhizophora
 (C) Nepenthes
 (D) Dracena
 The organic substance present in mesophyll cells are passed into the sieve tubes through their companion cells by
 - (A) an active transport (B) simple diffusion (C) facilitated diffusion (D) Osmosis
- 64. Guttation takes place through

62.

- (A) stomata (B) hydathodes (C) water pore (D) both (B) and (C)
- 65. Which enzyme of TCA/Krebs' cycle is not present in mitochondrial matrix?

An insectivorous angiosperm in which roots are absent is

- (A) Malate dehydrogenase (B) Citrate synthase
- (C) Aconitase (D) Succinate dehydrogenase
- 66. Which among the following is an inverted pyramid?
 - (A) Pyramid of energy in grassland (B) Pyramid of number in pond ecosystem
 - (C) Pyramid of number in grassland (D) Pyramid of biomass in an aquatic system
- 67. Match column I with column II and select the correct option

Col	umn I		Column II
1.	Golden rice	p.	High protein content
2.	Brassica napus	q.	Cry 1 Ab
3.	Bt corn	r.	β-carotene
4.	Transgenic potato	s.	Hirudin
(A)	1-a, 2-p, 3-r, 4-q	(B)	1-r, 2-s, 3-q, 4-p
(C)	1-q, 2-r, 3-p, 4-s	(D)	1-q, 2-s, 3-p, 4-r

- 68. Read the given statements.
 - (i) In prokaryotes, the photosynthetic pigments are found in the _____.
 - (ii) DCMU is a herbicide which blocks_____

Select the correct option which correctly fills the two blanks.

(i)
(A) Thylakoid
(B) Chloroplast
(C) Thylakoid
(D) Chloroplast
(E) Chloroplast
(C) Thylakoid
(C) PS II

- 69. The thylakoids of chloroplast are removed and kept in a culture medium containing carbon dioxide and water. If the set up is exposed to light, hexose sugars are not formed as end products. The most appropriate reason for this is that
 - (A) carbon assimilation cannot take place in the presence of light
 - (B) the pigment systems are not working
 - (C) the enzymes are not available
 - (D) the light trapping device is non-functional
- 70. What does the following cross represents?

$$\begin{array}{ccc} AA & \times & BB \\ \downarrow & & \\ & F_1 & \\ \downarrow Doubling \\ AAAABB & \end{array}$$

(A) Autopolyploidy

(B) Allopolyploidy

(C) Autoallopolyploidy

- (D) Spontaneous mutation
- 71. Match the organisms given in column I with their common names in column II and choose the correct option.

Column-I			Column-II
A. Chondrus		(i)	Peat moss
B. Sphagnum		(ii)	Spike moss
C. Cladonia		(iii)	Irish moss
D. Selaginella		(iv)	Reindeer moss
(A) A-(iii), B-(i), C-(iv), D-(ii) ((B)	A-(i	ii), B–(ii), C–(iv), D–(i)
(C) A-(iv), B-(i), C-(iii), D-(ii) ((D)	A–(i), B–(iv), C–(ii), D–(iii)

- 72. Pick out the wrong statements.
 - (i) The stamens in the Family Cucurbitaceae are synandrous, extrose and monothecous
 - (ii) The entire shoot is modified for assimilatory function in cladodes.
 - (iii) Adventitious roots of Pandanus is an example of clinging roots
 - (iv) Meristematic tissue is a group of thin-walled isodiametic cells which are capable of cell division.
 - (A) (i) and (ii) only
- (B) (iii) and (iv) only
- (C) (ii) and (iii) only
- (D) (i) and (iv) only

73. Study the given table.

Biomagnification

- It is the increase in concentration of non bio degradable substance in the food chain
- It is found in aquatic ecosystem only (ii)
- It does not result in organic loading (iii)
- It leads to toxicity in higher order consumers (iv) On the above given differences.
- (A) (i) and (ii) alone are correct
- (C) (ii) and (iv) alone are correct
- (B) (i) and (iii) alone are correct

It is found in oceans only

It leads to organic loading

plant nutrients

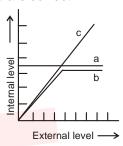
(D) (i), (iii), and (iv) alone are correct

Eutrophication

It is the enrichment of the water body with

It leads to toxicity in low order consumers

- Which of the following statements is incorrect?
 - (A) Birds and mammals belong to catagory 'a'.
 - (B) Animals in category 'b' always maintain constant internal environment.
 - (C) Osmotic concentration of animals in category 'c' changes according to ambient conditions.
 - (D) Animals of category 'b' shows suspended development during favourable conditions.



- Substrates used in floating respiration are 75.
 - (A) proteins only

(B) fats and proteins

(C) carbohydrates and fats

(D) carbohydrates only

Solanum tuberosum

- Bicarpellary syncarpous gynoecium is not found in the flowers of
 - (A) Atropa belladona
 - (D) Colchicum autumnale
- (C) Cestrum nocturnum 77. Spindle of plant cells is called
 - (A) amphiaster
- (B) anastral
- (C) acentric
- (D) both (B) and (C)
- Which of the following is/are autosomal recessive trait/traits 78.
 - (A) Albinism
- (B) Sickle cell anemia (C) Thalessemia
- (D) All of these

- 79. Which of the following statements is wrong?
 - (A) In Pinus, male gametophyte is free-living.
 - (B) The sporophyte of Polytrichum is more developed than that in Riccia.
 - (C) Wolffia is the smallest known angiosperm
 - (D) Salvinia is a heterosporous pteridophyte
- Select the option which clearly differentiates phellem from phelloderm 80.

Phellem

- (A) It is formed on the innerside of phellogen
- (B) It is made up of living cells
- (C) Its cells store the food
- (D) Suberisation is present

Phelloderm

- It is formed on the outerside of phellogen
- It is made up of dead cells
- It is protective in function
- Suberisation is absent



(B) Floating stage

(A) Submerged stage(C) Reed swamp stage

(D) March meadow stage

82. The double stranded DNA has 15% of cytosine. The percentage of adenine in DNA will be

(A) 35%

(B) 30%

(C) 45%

(D) 70%

[15]

83. Triticum is known as festucoid grass because

(A) vascular bundle is surrounded by single sheath of compactly arranged parenchyma cells

(B) vascular bundle is chlorenchymatous

(C) vascular bundle is sclerenchymatous

(D) vascular bundle is surrounded by double sheath of compactly arranged paranchyma cells

84. Lysozyme is not present in

(A) saliva

(B) tears

(C) tissue fluid

(D) sweat

85. Warm blooded animals of colder areas have large body size compared to animals of warmer areas. This is

(A) Allen's rule

(B) Gloger's rule

(C) Bergamann's rule

(D) Jordan's rule

86. Which of the following hormones is used to prevent the sprouting of potatoes?

(A) 2–4–dichlorophenoxy acetic acid

(B) 2, 4, 5-trichlorophenoxy acetic acid

(C) Indole-3-Acetic acid

(D) 2-methyl-4-chlorophenoxy acetic acid

(A) cytokinin, gibberellin

(B) auxin, cytokinin

(C) auxin, ethylene

(D) gibberellin, ethylene

88. Select the incorrect statement from the following.

(A) Water is absorbed by a system having DPD = 10 atm from another system having DPD= 5 atm

(B) The values of turgor pressure and solute potential are same in case of fully turgid cell.

(C) DPD becomes 0 in case of a flaccid cells.

(D) Osmotic potential is the reduction in free energy of water due to decrease in number of water molecules per molal volume.

89. Which of the following pairs is incorrectly matched?

(A) Kinetin – adenine derivative

(B) Gibberelline – terpenes (C) Ethylene – gases

(D) ABA – indole compounds

90. Flowers of Kigelia pinnata have abundant nectar and prominent stamens. They are pollinated by

(A) insects

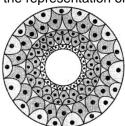
(B) winds

(C) birds

(D) bats



91. This diagrammatic sectional view is the representation of



- (A) Coelomate (B) Pseudocoelomate (C) Acoelemate (D) Both 2 and 3
- 92. True segmentation or metamerism means
 - (A) Body is externally and internally divided into segments
 - (B) Each segment of body have serial repetition of at least some organs
 - (C) Both (A) and (B)
 - (D) There is no repetition of any organ in successive segments.
- 93. Select the incorrect statement.
 - (A) In Urochordata, notochord is present only in larval tail.
 - (B) In Cephalochordate notochord is present in head region only and is persistent throughout
 - (C) In vertebrata, notochord is replaced by vertebral column in adult.
 - (D) All vertebrates are chordates but all chordates are not vertebrates.
- 94. Substrate concentration at which reaction attains half of its maximum velocity is called (A) $1/2 V_{max}$ (B) [S] 1/2 (C) K_{max} (D) K_{m}
 - (A) $1/2 V_{max}$ (B) [S] 1/2 (C) K_{max} Increased liquidity of the faecal discharge is known as
 - (A) Indigestion (B) Constipation (C) Diarrhoea (D) Vomiting
- 96. The causes of indigestion are
 - (A) Inadequate enzyme secretion (B) Over eating and spicy food
 - (C) Food poisoning and anxiety (D) All of these
- 97. A feeling of nausea before

95.

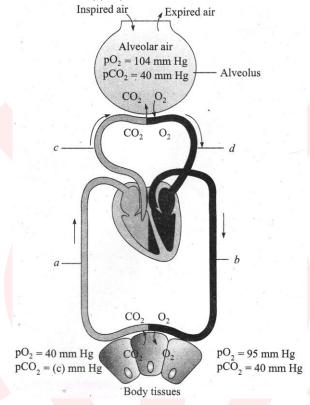
- (A) Indigestion (B) Constipation (C) Diarrhoea (D) Vomiting
- 98. Which of the following is not a correct matching group of developmental fates of the primary germ layers?
 - (A) Ectoderm : Epidermis, Central nervous system, sense organs, neural crest
 - (B) Mesoderm : Skeleton, muscles, blood vessels, heart, liver, gonads (C) Endoderm : Lining of digestive and respiratory tracts, pancreas
 - (D) Ectoderm : Pineal gland, pituitary gland, adrenal medulla
- 99. How much percent of O₂ and CO₂ is transported by the RBCs, respectively?
 - (A) 97%, 70% (B) 97%, 7% (C) 70%, 20–25% (D) 97%, 20–25%
- 100. How much percent of O_2 and CO_2 is carried in a dissolved state through the plasma, respectively?
- (A) 7%, 70% (B) 3%, 7% (C) 7%, 3% (D) 7%, 20–25%
- 101.a......contain a very high concentration of carbonic anhydrase and minute quantities of the same is present in the.......b.....too.

Fill in the blanks.

- (A) a-RBC, b-Plasma (B) a-Plasma, b-RBC
- (C) a–Hb, b–RBC (D) a–RBC, b–Serum



102. Recognize the figure and choose appropriate match.



- Pulmonary artery i.
- Systemic arteries iii.
- (A) a-iv, b-iii, c-i, d-ii
- (C) a-ii, b-i, c-iii, d-iv
- 103. T-Wave represents
 - (A) Depolarisation of atria

 - (C) Repolarisation
- 104. Excitation (Depolarisation) of the atria is represented by
 - (A) P-Wave
- (B) QRS-Complex
- (C) T-Wave

ii.

iv.

(B)

(D)

Pulmonary veins

a-iii, b-iv, c-i, d-ii

a-i, b-ii, c-iii, d-iv

Both (A) and (B)

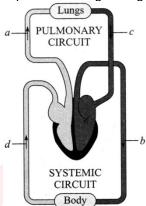
Depolarisation of Ventricles

Systemic veins

(D) Q-Wave

105. Recognize the figure and give appropriate labelling from given options.

Pulmonary artery



		••	· aminomany anto-	. ,			diii.idi.y . diii				
		iii.	Vena Cava				iv.	Dorsal A	orta		
		а	b		С			d			
		(A)	i	iii			ii			iv	
		(B)	i	iv			ii			iii	
		(C)	ii	iii			i			iv	
		(D)	ii	iv			i			iii	
1	06.	Norr	mal blood pressu	re is							
		(A)	120/80	(B)	80/120		(C)	140/80		(D)	140/70
1	07.	Prot	onephridia or flai	me cells a	e the exc	retory stru	icture:	s in			
		(A)	Planaria				(B)	Some Ro	otifers an	d Ann	elids
		(C)	Amphioxus				(D)	All of the	above		
1	08.	Nep	hridia are tubulai	r excretory	structure	s of					
		(A)	Earthworms	(B)	Cockroad	ches	(C)	Other An	nelids	(D)	Both (A) and (C)

ii.

Pulmonary Vein

Urinary bladder

(C) a, c and d

Ureter

(D) b and d

Space for Rough Work

109. Among the following which structure are unpaired in case of human excretory system

Urethra

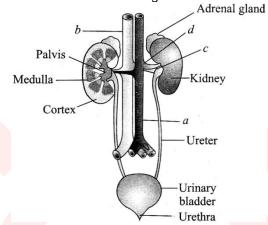
(B) b, c and d



Kidney

(A) b and c

110. Recognize the figure and find the correct labelling.



i.	Renal Artery	ii.	Renal vein	iii.	Inferior	ena c	ava iv. [Dorsal a	orta
	a	b		С			d		
(A)	iii	iv		i			ii		
(B)	iv	iii		i			ii		
(C)	iii	iv		ii			i		
(D)	iv	iii		ii			į		

- 111. Human skull is
 - (A) Monocondylic (B) Dicondylic (C) Tricondylic (D) Quadricondylic
- 112. Hyoid bone is.....a.....Shaped andb...
 - (A) a–C shaped, b–Paired (B) a–U shaped, b–Paired
 - (C) a-C shaped, b-Unpaired (D) a-U shaped, b-Unpaired
- 113. If the trophoblast layer failed to form in a mammalian embryo, which of the following structures would not develop?
 - (A) The blastopore (B) The inner cell mass
 - (C) The archenteron (D) The foetal placenta
- 114. The vertebral column of human is differentiated as the
 - (A) $C_7, T_{12}, L_5, S_5, Co_4$ (B) $C_7, T_{12}, L_5, S_4, Co_5$
 - $(C) \quad C_7, T_{12}, L_5, S_4, Co_1 \\ \qquad \qquad (D) \quad C_7, T_{12}, L_5, S_1, Co_1 \\$

						.ppcug/	. 455			[20]
115.	Whi	ch part of the bra	ain connec	ted to the spin	al cord?					
	(A)	Pons			(B)	Cerebellu	ım			
	(C)	Medulla			(D)	Cerebral	aqueduc	t		
116.	Holo	oblastic cleavage	e results in							
	(A)	Formation of a	symmetric	al blastula con	<mark>npos</mark> ed of c	ells of app	roximate	ly equ	ıal size	
	(B)	Formation of a	n asymme	trical blastula <mark>c</mark>	<mark>comp</mark> osed o	f cells of a	pproxima	ately ι	unequal	size
	(C)	Cell division of	cells only	near the anima	al pole					
	(D)	Cell division of	only the c	ells near the ve	egetal pole					
117.		ch contraceptiv smission of sexu				ting fertili	zation a	ınd p	rotectin	g against
	(A)	Oral contracep	tives		(B)	Diaphrag	ms			
	(C)	Nirodh			(D)	Intrauteri	ne device	e (IUD))	
118.	The	wall layer of eye	e ball whic	n looks bluish i	in <mark>colo</mark> ur					
	(A)	Sclera	(B)	Choroid	(C)	Cornea		(D)	Retina	l
119.	Whi	ch of the followir	ng structur	e forms t <mark>he bir</mark>	th canal?					
					a	d				
	(A)	a and b	(B)	b and c	(C)	c and d		(D)	a and	d
120.	Ova	ries are located	one on ea	ch side of						
	(A)	Lower pelvic re	aion		(B)	Lower ab	domen			

Space for Rough Work



(C) Upper abdomen

(D) Lower thoracic region