KELVIN IIT-JEE | MEDICAL | FOUNDATIONS

KELVIN Entrance Test (KET)

Class : 12th Passed (Engineering)

Code : A

Time : 80 minutes		MM : 171
Registration Number	:	
Name of the Candidate	:	
Test Centre	:	

Instructions :

Caution : Class, Paper, Code as given above must be correctly marked in the answer OMR sheet before attempting the paper. Wrong Class, Paper or Code will give wrong results.

1. This question paper consists of 45 questions. All questions will be multiple choice single correct out of four choices with marking scheme in table below.

Subject	Question No.	Marking scheme for each question		
		Correct Answer	Incorrect Answer	
PHYSICS	Q. 1-6	+3	0	
	Q. 7-12	+4	0	
	Q. 13-15	+5	0	
CHEMISTRY	Q. 16-21	+3	0	
	Q. 22-27	+4	0	
	Q. 28-30	+5	0	
MATHEMATICS	Q. 31-36	+3	0	
	Q. 37-42	+4	0	
	Q. 43-45	+5	0	

- 2. Answers have to be marked on the OMR sheet. The Question Paper contains blank spaces for your rough work. No additional sheets will be provided for rough work.
- 3. Blank papers, clip boards, log tables, slide rule, calculator, cellular phones, pagers and electronic devices, in any form, are not allowed.

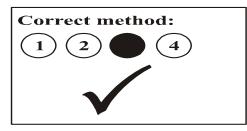
- 4. Before attempting paper write your Registration Number, Name and Test Centre in the space provided at the top of this sheet.
- 5. See method of marking of bubbles of the back of cover page for question no. 1 to 45.

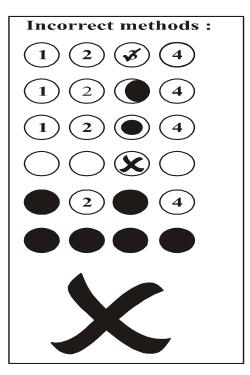
Note : Please check this Question Paper contains all 45 questions in serial order. If not so, exchange for the correct Question Paper.

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Method of marking of bubbles for questions number 1 to 45. For example : Question number 19 :

If correct option is 3, then





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PHYSICS

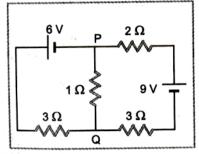
1. A particle of unit mass undergoes one – dimensional motion such that its velocity varies according to $v(x) = \beta x^{-2n}$, where β and *n* are constant and *x* is the position of the particle. The acceleration of the particle as a function of *x*, is given by

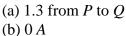
(a) $-2 n \beta^2 x^{-4n-1}$ (b) $-2 n \beta^2 x^{-n+1}$ (c) $-2 n \beta^2 x^{-4n+1}$ (d) $-2 n \beta^2 x^{-2n-1}$

- **2.** The moment of inertia of a solid sphere of density *ρ* and radius *R* about its diameter is :
 - (a) $\frac{105}{176}R^5\rho$ (b) $\frac{176}{105}R^5\rho$ (c) $\frac{105}{176}R^2\rho$ (d) $\frac{176}{105}R^2\rho$
- 3. Charge Q is distributed to two different metallic spheres having radii R and 2 R such that both spheres have equal surface charge density. Then charge on large sphere is

(a) $\frac{4Q}{5}$			(b) $\frac{q}{5}$
			$(0)\frac{1}{5}$
$(c)\frac{3Q}{5}$			$(d) \frac{50}{5}$
$(0) \frac{1}{5}$			$(u) - \frac{1}{5}$
т .1	• •,	1	.1

4. In the circuit shown, the current in the 1 Ω resistor is Fig





(c) 0.13 A from Q to P

(d) 0.13 *A*, from *P* to *Q*

5. The refracting angle of a prism is A, the refractive index of the material of the prism is

 $\cot\left(\frac{A}{2}\right)$. The angle of minimum deviation is:

- (a) $180^\circ 2A$ (b) $90^\circ A$
- (c) $180^\circ + 2A$ (d) $180^\circ 3A$
- 6. A radioactive nucleus (initial mass number A and atomic number Z) emits 3α particles and 2 positrons. The ratio of number of neutrons to the of proton in the final nucleus will be

(a)
$$\frac{A-Z-4}{Z-8}$$
 (b) $\frac{A-Z-12}{Z-4}$
(c) $\frac{A-Z-4}{Z-2}$ (d) $\frac{A-Z-8}{Z-4}$

7. A bomb at rest explodes into three parts of the same mass. The moments of the two parts are -2 p î and p ĵ. The momentum of the third part will have a magnitude of :

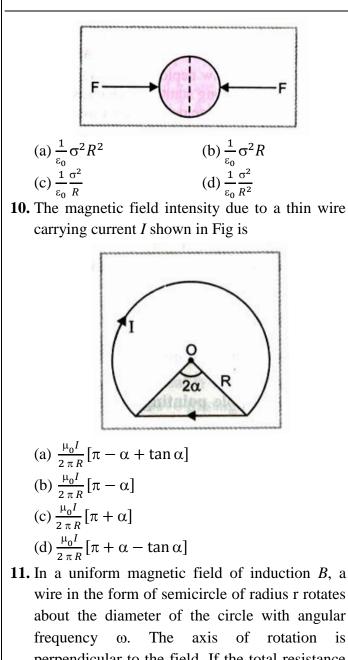
(a) p (b)
$$\sqrt{3p}$$

(c)
$$p\sqrt{5}$$
 (d) zero

8. From a solid sphere of mass M and radius R, a cube of maximum possible volume is cut. Moment of inertia of cube about an axis passing through its centre and perpendicular to one of its faces is

(a)
$$\frac{MR^2}{32\sqrt{2}\pi}$$
 (b) $\frac{MR^2}{16\sqrt{2}\pi}$
(c) $\frac{4MR^2}{9\sqrt{3}\pi}$ (d) $\frac{4MR^2}{3\sqrt{2}\pi}$

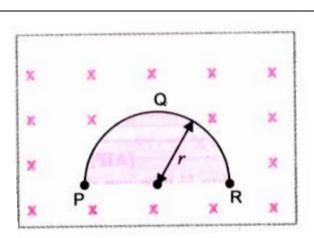
9. A uniformly charged thin spherical shell of radius R carries uniform surface charge density of σ per unit area. It is made of two hemispherical shells, held together by pressing them with force *F*, fig is proportional to



perpendicular to the field. If the total resistance of the circuit is R, then the mean power generated per period of rotation is

(a)
$$\frac{B \pi r^2 \omega}{2 R}$$
 (b) $\frac{(B \pi r^2 \omega)^2}{16 R}$
(c) $\frac{(B \pi r \omega)}{2 R}$ (d) $\frac{(B \pi r^2 \omega)^2}{8 R}$

12. A thin semicircular conducting ring (PQR) of radius r is falling with its plane vertical in a horizontal magnetic field B, as shown in Fig The potential difference developed across the ring its speed is v, is



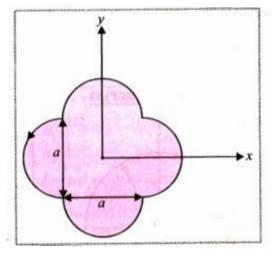
(a) Zero

- (b) $B v \pi r^2/2$ and P is at higher potential
- (c) $\pi r B v$ and R is at higher potential
- (d) 2 r Bv and R is at higher potential.
- 13. The densities of two sphere A and B of same radii R vary with radial distance r as $\rho_A(r) =$

 $K\left(\frac{r}{R}\right)$ and $\rho_B(r) = K\left[\frac{r}{R}\right]^5$. respectively,

where *k* is a constant. The moments of inertia of the individual spheres about axes passing through their centers are I_A and I_B respectively. If $\frac{I_B}{I_A} = \frac{n}{10}$, the value of *n* is

- (a) 6
- (b) 10
- (c) 16
- (d) 7
- 14. A loop carrying current I lies in the *x*-*y* plane as shown in the Fig The unit vector \hat{k} is coming out of the plane of the paper. The magnetic moment of the current loop is



 (a) a²I k (b) (π/2 + 1) a²I k (c) - (π/2 + 1) a²I k (d) (2π + 1)a²I k 15. The focal length of a thin biconvex lens is 20 cm. When an object is moved from a distance of 25 cm in front of it to 50 cm, the magnification 	of its image changes from m_{25} to m_{50} . The ratio $\frac{m_{25}}{m_{50}}$ is (a) 4 (b) 6 (c) 1 (d) 3
	EMISTRY
16. Which of the following molecules is planar? (a) SF ₄ (b) XeF ₄ (c) NF ₃ (d) SiF ₄ 17. The number of electrons involved in the conversion of MnO ₄ to MnO ₂ is (a) 3 (b) 4 (c) 1 (d) 2 18. Two isomers of a compound Co(NH ₃) ₃ Cl ₃ (MA ₃ B ₃ type) are shown in the figures. $H_{3N} \longrightarrow Cl$ $H_{3N} \longrightarrow Cl$ $H_{3N} \longrightarrow Cl$ $H_{3N} \longrightarrow Cl$ $H_{3N} \longrightarrow Cl$ $H_{3N} \longrightarrow Cl$ The isomers can be classified as (a) (i) fac-isomers (ii) mer-isomer (b) (i) optical-isomer (ii) trans-isomer (c) (i) mer-isomer (ii) fac-isomers (d) (i) trans-isomer (ii) cis-isomer. H3. CH ₃ CH ₂ CH ₂ Br $\longrightarrow CH_3$ CH=CH ₂ The above reaction is an example of reaction. (a) substitution (b) elimination (c) addition (d) rearrangement 20. Paracetamol is used as an (a) antibiotic (b) antimalarial (c) antipyretic (d) arsenical 21. Correct IUPAC name for $H_3C-CH-CH-CH_3$ is $L = \frac{1}{C_2H_5} C_{2H_5}$ (a) 2-Ethyl-3-methylpentane (b) 3, 4-Dimethylpentane (c) 2-sec-Butylbutane (d) 2, 3-Dimethyl butane	 22. The ratio of kinetic energy and potential energy of an electron in any orbit is equal to (a) 0 (b) -1/2 (c) -2 (d) ∞ 23. Maximum enol content is in: (a) √ (b) √ (c) -2 (c) -2 (d) √ (e) √ (f) √ (f) √ (g) √ (h) √ (h) √ (g) ∞ 24. Relative lowering of vapour pressure of a dilute solution is 0.2. What is the mole fraction of the non-volatile solute? (a) 0.8 (b) 0.5 (c) 0.3 (d) 0.2 25. For the reaction 4NH₃ + 5O₂ → 4NO + 6H₂O, if the rate of disappearance of NH₃ is 3.6 × 10⁻³ mol L⁻¹s⁻¹ (b) 3.6 × 10⁻³ mol L⁻¹s⁻¹ (c) 4 × 10⁻³ mol L⁻¹s⁻¹ (d) 0.6 × 10⁻⁴ mol L⁻¹s⁻¹ (e) 4 × 10⁻⁴ mol L⁻¹s⁻¹ (f) 0.6 × 10⁻⁴ mol L⁻¹s⁻¹ (g) 0.6 × 10⁻⁴ mol L⁻¹s⁻¹ (h) solid is dispersed in a liquid (c) gas is dispersed in a liquid (d) sugar is dispersed in a liquid (e) sugar is dispersed in a liquid (f) Ar (g) Ne (g) He 28. When a substance (A) react with water it produces a combustible gas (B) and a solution

	10
of substance (C) in water. When another	alcoholic group is isotopically labeled with O ¹⁸
substance (D) reacts with this solution of (C), it	as shown. Identify the products when it is
also produces the same gas (B) on warming but	dehydrated with H ₃ PO ₄ .
(D) can produce gas (B) on reaction with dilute	
sulphuric acid at room temperature. Substance	
	$\xrightarrow{H_3PO_4}$
(A) imparts a deep golden yellow colour to a	HO_{18} OH
smokeless flame of Bunsen burner. Then (A),	No On
(B), (C) and (D) respectively are	
(a) CaH_2 , $Ca(OH)_2$, Sn	(a) (18) $+H_2O$
(b) K, H ₂ , KOH, Al	
(c) Na, H ₂ , NaOH, Zn	
(d) CaC_2 , C_2H_2 , $Ca(OH)_2$, Fe	
29. NaCN is sometime added in the froth floatation	0
process as a depressant, when ZnS and PbS	(b) $\xrightarrow{0}$ $\xrightarrow{18}$
	(0) $/$ $+$ H_2O
mineral are extracted because	
(a) Zns forms soluble complex $Na_2[Zn(CN)_4]$	
while PbS forms froth	
(b) $Pb(CN)_2$ is precipitated while no effect on	(c) $+H_2O^{18}$
ZnS	
(c) PbS forms soluble complex $Na_2[Pb(CN)_4]$	\sim
while ZnS forms froth	0
(d) NaCN is never added in froth floatation	10 18
	(d) $/$ $+$ H_2O
process	\sim
30. Consider the following diol which has one 2°	
alcoholic group and one 3° alcoholic group. 3°	
MATH	EMATICS
31. For real number x and y, we write $xRy \Leftrightarrow x - y$	(c) $\frac{\sqrt{5}+1}{5}$ (d) $\frac{\sqrt{5}+1}{2\sqrt{2}}$
$+\sqrt{2}$ is an irrational number. Then the relation	
R is	35. $\cos(\tan^{-1}x) = 1$
(a) Reflexive (b) Symmetric	(a) $\sqrt{1+x^2}$ (b) $\frac{1}{\sqrt{1+x^2}}$
(c) Transitive (d) None of these	(c) $1 + x^2$ (d) None of these
32. The interior angles of polygon are in A.P if the	36. The area of the triangle, whose vertices (1, 0),
smallest angle be 120° and the common	(7, 0) and (4, 4) is
difference be 5°, then the number of sides of the	(a) 8 (b) 10
polygon	(c) 12 (d) 14
(a) 8 (b) 10	37. The equation of line through (3, -4) and
(c) 9 (d) 6	perpendicular to the line $3x + 4y = 5$ is
33. If $A = \begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \end{bmatrix}$, then $A^2 =$	(a) $4x + 3y = 24$
33. If $A = \begin{bmatrix} 0 & 1 & 0 \\ 1 & 0 \end{bmatrix}$, then $A^2 =$	(b) $y - 4 = (x + 3)$
La b -1J (a) Unit matrix (b) Null matrix	(c) $3y - 4x = 24$
$\begin{array}{c} (a) \text{ Unit matrix} \\ (b) \text{ Null matrix} \\ (c) \text{ A} \\ (d) \text{ -A} \end{array}$	(d) $y + 4 = \frac{4}{3}(x - 3)$
34. The value of $\cos^2 48^\circ - \sin^2 12^\circ$ is	38. If the line $\ddot{x} + 2by + 7 = 0$ is a diameter of the
	circle $x^2 + y^2 - 6x + 2y = 0$, then b =
(a) $\frac{\sqrt{5}+1}{8}$ (b) $\frac{\sqrt{5}-1}{8}$	(a) 3 (b) -5
	(c) -1 (d) 5

-0-

39. The directrix of the hyperbola is $\frac{x^2}{9} - \frac{y^2}{4} = 1$ (a) $x = 9/\sqrt{13}$ (b) $y = 9/\sqrt{13}$ (c) $x = 6/\sqrt{13}$ (d) $y = 6/\sqrt{13}$ **40.** If the vectors $2\hat{i} - 3\hat{j}$, $\hat{i} + \hat{j} - \hat{k}$ and $3\hat{i} - \hat{k}$ form three concurrent edges of a parallelopiped, then the volume of the parallelopiped is (a) 8 (b) 10 (d) 14 (c) 4 **41.** The shortest distance between the lines $\frac{x-3}{3} =$ $\frac{y-8}{-1} = \frac{z-3}{1}$ and $\frac{x+3}{-3} = \frac{y+7}{2} = \frac{z-6}{4}$ is (b) $2\sqrt{30}$ (a) $\sqrt{30}$ (c) $5\sqrt{30}$ (d) $3\sqrt{30}$ 42. $\lim_{\theta \to 0} \frac{\tan x - \sin x}{x^3} =$ (a) 1/2(b) -1/2

(c) 2/3 (d) none of these 43. Let $h(x) = f(x) - (f(x))^2 + (f(x))^3$ for every real number x. then (a) h is increasing whenever f is increasing (b) h is increasing whenever f is decreasing (c) h is decreasing whenever f is increasing (d) Nothing can be said in general 44. If $I_n = \int_0^{\pi/4} \tan^n \theta \, d\theta$, then $I_8 + I_6$ equals (a) $\frac{1}{4}$ (b) $\frac{1}{5}$ (c) $\frac{1}{6}$ (d) $\frac{1}{7}$ 45. The value of $\int_{\sqrt{In2}}^{\sqrt{In3}} \frac{x \sin x^2}{\sin x^2 + \sin(In6 - x^2)} \, dx$ is (a) $\frac{1}{4} \ln \frac{3}{2}$ (b) $\frac{1}{2} \ln \frac{3}{2}$ (c) $\ln \frac{3}{2}$ (d) $\frac{1}{6} \ln \frac{3}{2}$

ANSWER KEY

PHYSICS

1	Α	11	D
2	В	12	D
3	Α	13	Α
4	С	14	В
5	Α	15	В
6	Α		
7	С		
8	С		
9	Α		
10	Α		

CHEMISTRY

16	B	26	A
17	Α	27	D
18	Α	28	С
19	В	29	Α
20	С	30	В
21	В		
22	В		
23	В		
24	D		
25	A		

MATHEMATICS

		1	1
31	Α	41	D
32	С	42	Α
33	Α	43	Α
34	Α	44	D
35	В	45	Α
36	С		
37	D		
38	D		
39	Α		
40	С		