

**B. CHEMICAL ENGINEERING FINAL EXAMINATION, 2008**

( 2nd Semester, Supplementary )

**CHEMICAL PROCESS SYNTHESIS**

Time : Three hours

Full Marks : 100

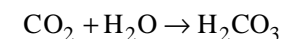
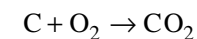
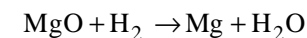
Answer question *no. 3* and any *five* from the rest

1. There is a hot stream  $Sh_1$ , in a chemical plant with an inlet temperature of  $400^\circ\text{C}$  and heat capacity flow rate of 100 units to be cooled to  $200^\circ\text{C}$ . In the same chemical plant, there are two more cold streams  $Sc_1$  and  $Sc_2$  with their inlet temperature as  $100^\circ\text{C}$  and  $200^\circ\text{C}$  respectively to be heated to  $150^\circ\text{C}$  and  $250^\circ\text{C}$  respectively. Heat capacity flow rates  $Sc_1$  and  $Sc_2$  are assumed to be 200 units each.

Show different ways in which the task can be achieved. Show those with the help of diagrams.

Show also that the order of contacting hot and cold streams is extremely important. 6+10

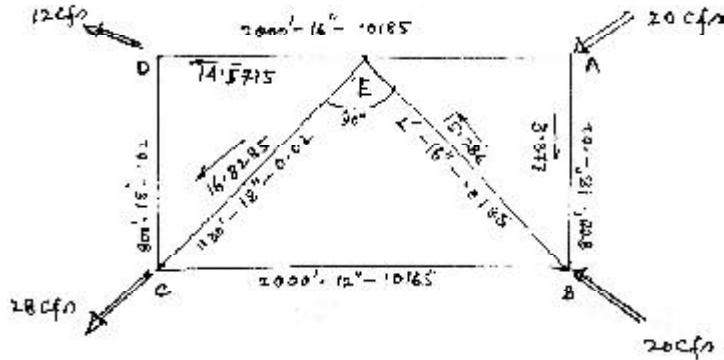
2. a) State the deduction theorem of propositional logic and prove it.
- b)  $\text{MgO}$ ,  $\text{H}_2$ ,  $\text{O}_2$  and  $\text{C}$  are raw materials and can perform the following reactions.



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Draw a flow sheet for the production of  $H_2CO_3$  using deduction theorem of propositional logic. 4+12

3.



Water is flowing through network shown above. This network is to be analysed using Hardy Cross procedure. Length of the pipe in ft, inside diameter of the pipe in inch and fanning friction coefficient are shown against each pipe line in order. Length L is not known. Find the flowrate and pressure drop in each pipe line using Hardy Cross method (complete one iteration and find the corrected flow rate in each member. Use this corrected flowrate for pressure drop calculation). Initial guess of flow rate at some members are suggested in the figure. 20

4. Draw an optimal separation flowsheet to achieve goals  $G_1$ ,  $G_2$  &  $G_3$ , when two streams  $S_1$ ,  $S_2$  are available. The order of achieving the goal is  $G_2$ ,  $G_1$ ,  $G_3$ .

7. Write short notes on the following (*any four*)

- i) Guide rule for estimating split fraction coefficients for absorption or stripping column.
- ii) FLOWTRAN
- iii) Manning equation for pipe friction.
- iv) Linearization method for analysis of pipe line network.
- v) Main constituent features of full steady state simulation programs for computer aided flowsheeting. 454

6. Rawmaterials and reaction list is given below and the goal G is a mixture of  $A_6$  and  $B_7$ . Draw a steady state flow diagram for the purpose.

16

Raw mat.	Species (liq)	Temp °C	Press atm	Reaction	Input	Output	Temp, °C	Press atm
S <sub>1</sub>	A <sub>5</sub> , B <sub>4</sub>	30	2	██████████	A <sub>1</sub> , A <sub>2</sub>	A <sub>1</sub> , A <sub>3</sub> , A <sub>5</sub>	100	2
S <sub>2</sub>	A <sub>2</sub> , B <sub>3</sub>	25	1	██████████	A <sub>3</sub> , A <sub>5</sub>	(i) A <sub>6</sub> , A <sub>5</sub> (liquid)	90	1
S <sub>3</sub>	A <sub>1</sub> , B <sub>1</sub> , B <sub>2</sub>	25	1	██████████	A <sub>3</sub> , A <sub>8</sub>	(ii) A <sub>7</sub> , A <sub>5</sub> (gas)	50	1
S <sub>4</sub>	A <sub>8</sub> , A <sub>9</sub>	30	1					
S <sub>5</sub>	A <sub>11</sub> , B <sub>7</sub>	25	1	██████████				

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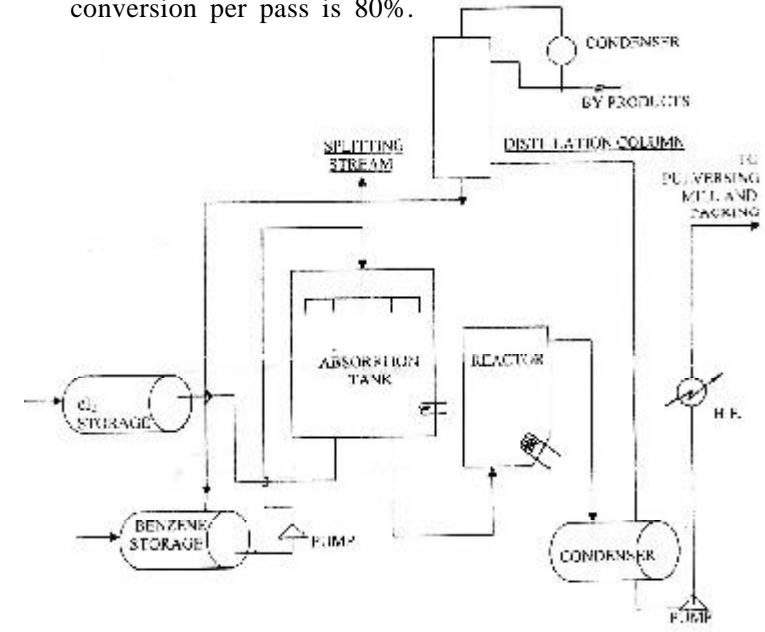


Figure (1)  
FLOW SHEET FOR MANUFACTURING BENZENE HEXACHLORIDE

Stream	X <sub>A</sub>	X <sub>B</sub>
S <sub>1</sub>	3	7
S <sub>2</sub>	7	3
G <sub>1</sub>	2	0
G <sub>2</sub>	2	7
G <sub>3</sub>	6	3

Use Mortard rule.

5. Develop the material balance equations afterputting the split fraction coefficients. Show the equations in matrix form and write the first estimate of the split fraction coefficients with reasoning. (Fig(1) may be refered. The yield is 60% and conversion per pass is 80%.

16

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