## SEAL



Test Booklet Number

Subject Code - 1201

Roll Number

00773

**MATHEMATICS & SCIENCE** 

[Maximum Marks: 300]

[Time: 2 Hours]

## **INSTRUCTIONS TO CANDIDATES**

Read the following instructions carefully before you answer the questions given in this Test Booklet:

- 1. Answers to questions in this Test Booklet are to be given on an OMR Answer Sheet provided to the candidate separately.
- 2. Candidate must fill up Name, Category, Test Booklet Number, Subject Code and Roll Number in the Answer Sheet carefully as per instructions given.
- 3. This Test Booklet consists of 75 questions. All questions are compulsory and carry equal marks.
- 4. Each question in this Test Booklet has four possible alternative answers namely, (A), (B), (C) and (D), one of which is correct. Candidate should choose the correct answer against each question out of four alternative answers.
- Candidate is instructed to answer the questions by darkening ( ) with Ball Point Pen only in the circle bearing the correct answer.
- 6 Candidate should not attempt more than one answer in each question. More than one attempt in any form against a question shall be treated as incorrect.
- 7. Marking of answer other than darkening shall be cancelled and darkening should remain within the circle or otherwise computer shall not accept during evaluation of answer-script.
- 8. Rough work must not be done on the Answer Sheet. Use the blank space given in the Test Booklet for rough work.
- 9. Candidate is to hand over the Answer Sheet to the Invigilator before leaving the Examination Hall.
- 10. **NEGATIVE MARKING**: Each question carries 4 (four) marks for correct response. For each incorrect response, 1 (one) mark will be deducted from the total score. More than one answer indicated against a question will be deemed as incorrect response and will be negatively marked.

P.T.O.



## **MATHEMATICS**

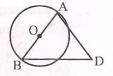
- 1. The irrational number between 2 and 3 is
  - (A)  $\sqrt{2}$
  - (B)  $\sqrt{3}$
  - (C)  $\sqrt{5}$
  - (D)  $\sqrt{11}$
- 2. If  $\sqrt{6} = 2.448$ , then the value of  $\frac{3\sqrt{2}}{2\sqrt{3}}$  is
  - (A) 0.816
  - (B) 1.224
  - (C) 0.613
  - (D) 2.449
- 3. The remainder, when  $p(x)=x^4+2x^3-3x^2$ + x-1 is divided by x+2 is
  - (A) -21
  - (B) 21
  - (C) 15
  - (D) -15
- 4. In the given figure, AB || CD, then the value of x is
  - (A) 80°

(B)

880

- E x 130°
- (C) 90°
- (D) 98°
- 5. If ABCD is an isosceles trapezium with AB $\parallel$ DC, then  $\angle$ C is equal to
  - (A) ∠B
  - (B) ∠A
  - (C) ∠D
  - (D) 90°

- 6. In given figure, AB is a diameter of circle with centre O. D is a point lying outside the circle. Then ∠ADB is
  - (A) 90°
  - (B) greater than 90°
  - (C) less than 90°



- (D) None of these
- 7. The perimeter of a triangular field is 144 m and the ratio of its sides is 3:4:5. The area of the field is
  - (A)  $864 \text{ m}^2$
  - (B)  $468 \text{ m}^2$
  - (C)  $824 \text{ m}^2$
  - (D)  $1440 \text{ m}^2$
- 8. If the height of a cube is equal to the diameter of a sphere, then taking  $\pi = \frac{22}{7}$ , the ratio of the volume of the cube to that of the sphere is
  - (A) 3:4
  - (B) 21:11
  - (C) 4:3
  - (D) 11:21
- 9. The mean of first five prime numbers is
  - (A) 5.6
  - (B) 3.6
  - (C) 6.83
  - (D) 5.2

D

10. A car is going on a long journey for 16 hours, starting at 5.00 hrs. The speeds of the car at different hours are as under:

Time (in hr) :	5	7	9	11	13	15	17	19	21
Speed (in kmph):	40	50	60	80	70	65	75	60	50

The percentage increase in speed during 9.00 hrs and 11.00 hrs is

- (A)  $33\frac{1}{3}\%$
- (B) 35%
- (C)  $23\frac{1}{3}\%$
- (D) 45%
- 11. If  $\alpha$ ,  $\beta$  are the zeroes of the polynomial p(x)

$$=4x^2+4x-1$$
, then  $\frac{1}{\alpha}+\frac{1}{\beta}$  is equal to

- (A) 4
- (B) -4
- (C)  $\frac{1}{4}$
- (D)  $-\frac{1}{4}$
- 12. If the pair of linear equations 2x + 3y = 11 and (m n) x + (2m 4n) y = 22, has infinitely many solutions, then
  - (A) m = 1, n = 5
  - (B) m = -1, n = 5
  - (C) m = 5, n = -1
  - (D) m = 5, n = 1

- 13. The roots of the quadratic equation  $3x^2-kx+14=0$  are in the ratio 7:6, then the value of k is
  - (A) -3
  - (B) 1
  - (C) 12
  - (D) 13
  - 14. The value of x for which the terms 2x, x + 10 and 3x + 2 are in A.P., is
    - (A) 2
    - (B) 4
    - (C) 6
    - (D)
- 15. In a ΔABC, D and E are points on sides AB and AC respectively such that BCED is a trapezium. If DE: BC = 3:5, then

$$\frac{\text{area }(\Delta ADE)}{\text{area }(\text{Trap.BCED})} \text{ is equal to}$$

- (A)  $\frac{3}{4}$
- (B)  $\frac{9}{16}$
- (C)  $\frac{3}{5}$
- (D)  $\frac{9}{25}$

- 16. The value of k for which (-3, 12), (7, 6) and (k, 9) are collinear, is
  - (A) 3
  - (B) 4
  - (C) 2
  - (D) 1
- 17. If  $\theta$  is acute and  $\frac{\cos^2 \theta}{\cot^2 \theta \cos^2 \theta} = 3$ , then the value of  $\theta$  is
  - (A) 30°
  - (B) 45°
  - (C) 90°
  - (D) 60°
- 18. In a right triangle ABC, right angled at B, BC = 6 cm and AB = 8 cm. A circle is inscribed in  $\triangle$ ABC. The radius of this circle is
  - (A) 1 cm
  - (B) 2 cm
  - (C) 3 cm
  - (D) 4 cm

- 19. The diameter of a copper sphere is 6 cm. It is melted and drawn into a wire of diameter 0.2 cm. The length of wire is
  - (A) 36 cm
  - (B) 360 cm
  - (C) 3600 cm
  - (D) 360 m
- 20. The probability that in a family of 3 children, there will be at least one boy, is
  - (A)  $\frac{7}{8}$
  - (B)  $\frac{1}{8}$
  - (C)  $\frac{3}{8}$
  - (D)  $\frac{6}{8}$

## SCIENCE

- 21. The angular velocity of the minutes hand of a clock is
  - (A)  $\frac{\pi}{180}$  rad/s
  - (B)  $\frac{\pi}{1800}$  rad/s
  - (C)  $\frac{\pi}{60}$  rad/s
  - (D)  $\frac{\pi}{360}$  rad/s
- 22. Action reaction forces
  - (A) act on the same body
  - (B) act on different bodies
  - (C) act along different lines
  - (D) act in the same direction
- 23. A force of 16 N is distributed uniformly on one surface of a cube of edge 8 cm. The pressure on this surface is
  - (A) 3500 Pa
  - (B) 2500 Pa
  - (C) 4500 Pa
  - (D) 5500 Pa

- 24. The weight of a body is measured to be 120 N on the earth. If it is taken to the moon, its weight will be about
  - (A) 120 N
  - (B) 60 N
  - (C) 20 N
  - (D) 720 N
- 25. A body at rest can have
  - (A) speed
  - (B) velocity
  - (C) momentum
  - (D) energy
- 26. Longitudinal waves cannot travel through
  - (A) vacuum
  - (B) solids
  - (C) liquids
  - (D) gases

- 27. The time period of a simple pendulum is1.2 second. If the length of the pendulum isdoubled, the new time period will be
  - (A) 1.1 s
  - (B) 1.3 s
  - (C) 1.5 s
  - (D) 1.7 s
- 28. The focal length of a concave mirror is f and the distance of the object to the principal focus is p. Then the ratio of the size of the image to the size of the object is
  - (A)  $\frac{f}{p}$
  - (B)  $\frac{p}{f}$
  - (C) fp
  - (D)  $\sqrt{\text{fp}}$
- 29. The property of an optical instrument, by virtue of which it can form separate images of two close objects, is
  - (A) magnifying power
  - (B) dispersive power
  - (C) resolving power
  - (D) illumination
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- 30. One cannot see through fog because
  - (A) fog absorbs light
  - (B) refractive index of fog is unity
  - (C) light suffers total internal reflection at the droplets in fog
  - (D) light is scattered by the droplets in fog
- 31. A ray of light incident normally on one face of prism just suffers total internal reflection at the other face. The refracting angle of the prism is
  - (A) 30°
  - (B) 45°
  - (C) 60°
  - (D) 75°
- 32. Electric lines of force about a negative charge are
  - (A) circular and anti-clockwise
  - (B) circular and clockwise
  - (C) radial and inward
  - (D) radial and outward

7

33.	How much electrical energy in kilowatt hours	s 36.	At gr	rid substation, the voltage is stepped up
	is consumed in operating ten 50-watt bulbs	S	to re	duce loss of
	for ten hours per day in a month of 30 days?  (A) 1500	?	(A)	current
	(A) 1500 (B) 15000		(B)	electrical energy
	(C) 15		(C)	power
	(D) 150		(D)	resistance
34.	An electrician has only two resistances. By	37.	Whe	en steam is passed over hot coke, it
	joining them in series or in paralle	1	prod	uces
	combinations, 3, 4, 12 and 16 $\Omega$ resistances can be obtained. The two resistances are	S	(A)	producer gas
	(A) $4 \Omega$ and $16 \Omega$		(B)	water gas
	(B) $4 \Omega$ and $12 \Omega$		(C)	laughing gas
	(C) $7 \Omega$ and $9 \Omega$		(D)	coal gas
	(D) $6 \Omega$ and $10 \Omega$	38.	Cho	ose the source of energy which is
35.	A compass needle just above a wire in which	1	diffe	rent from others
	electrons are moving towards east, will poin	t .	(A)	Natural gas
	(A) east (B) west		(B)	Petroleum
	(C) north has been (C)		(C)	Uranium
	(D) south both links (C)		(D)	Wood
1201	/Math & Sc.	8		Contd.

39.	Which of the following is the biggest planet?	42. Identify the pair which is not isoelectronic
	(A) Jupiter	from the following options
	(B) Venus	(A) Mg <sup>2+</sup> , F <sup>-</sup> (B) O <sup>2-</sup> , N <sup>3-</sup>
	(C) Mercury	(C) Na, Mg <sup>2+</sup>
	(D) Pluto	(D) $Li^+, Be^{2+}$
40.	The outermost layer of the earth is called	43. The element of atomic number 12 is closest in chemical properties of the element having
	(A) Crust	atomic number
	(B) Mantle	(A) 11 (B) 13
	(C) Core	(C) 20 mod distribution
	(D) Organic complex	(D) 18
41.	Which one of the following was the correct condition for carrying out the gas discharge	44. Oxidation of ammonia takes place in the following manner:
	tube experiment for the discovery of electrons?	$4NH_3(g) + 5O_2(g) \rightarrow 6H_2O(g) + 4NO(g)$ The mass of water that can be produced from 34 g of ammonia after oxidation will
	(A) High pressure, high voltage	be (molecular mass of ammonia = 17)
	(B) Low pressure, low voltage	(A) 9.0 g (B) 18.0 g
	(C) Low pressure, high voltage	(C) 36.0 g
	(D) High pressure, high temperature	(D) 54 0 g

- 45. What name is given to the reaction between hydrogen ion (H<sup>+</sup>) and hydroxyl ion (OH<sup>-</sup>)?
  - (A) Hydrolysis
  - (B) Neutralization
  - (C) Ionization
  - (D) Hydrogenation
- 46. Which one of the following reactions involves oxidation and reduction?
  - (A)  $Na_2SO_4(aq) + BaCl_2(aq) \rightarrow$  $NaCl(aq) + BaSO_4(s)$
  - (B)  $Zn(s) + H_2SO_4(aq) \rightarrow ZnSO_4(aq) + H_2(g)$
  - (C) NaCl(aq) + AgNO<sub>3</sub>(aq)  $\rightarrow$  AgCl(s) + NaNO<sub>3</sub>(aq)
  - (D)  $A1(NO_3)_3(aq) + 3NH_4OH(aq) \rightarrow$  $A1(OH)_3(s) + 3NH_4NO_3(aq)$

- 47. An element A with atomic number 11 combines with another element B having atomic number 17 to form a solid C. The solid C
  - (A) has high melting point and high boiling point
  - (B) has low melting point and low boiling point
  - (C) is good conductor of electricity
  - (D) is insoluble in water
- 48. Which one of the following salts will form basic solution on dissolving in water?
  - (A) Na<sub>2</sub>CO<sub>3</sub>
  - (B) Na<sub>2</sub>SO<sub>4</sub>
  - (C) NaCl
  - (D) KCI
- 49. Which one of the following metals cannot be obtained by reduction using carbon?
  - (A) Fe
  - (B) Zn
  - (C) Cu
  - (D) Mg

- 50. The reducing agent used in thermite process is
  - (A) carbon
  - (B) hydrogen
  - (C) carbon monoxide
  - (D) aluminium powder
- 51. Plaster of Paris is represented as
  - (A) CaSO<sub>4</sub> .2H<sub>2</sub>O
  - (B)  $CaSO_4.\frac{1}{2}H_2O$
  - (C) CaCO<sub>3</sub>.CaSO<sub>4</sub>
  - (D) CaSO<sub>4</sub>.3H<sub>2</sub>O
- 52. Froth floation technique is generally used to concentrate
  - (A) oxide ores
  - (B) sulphide ores
  - (C) carbonate ores
  - (D) halide ores

- - (A) 6 g
  - (B) 9 g
  - (C) 12 g
  - (D) 15 g
- 54. Chemical formula of baking soda is
  - (A) Na<sub>2</sub>CO<sub>3</sub>.10H<sub>2</sub>O
  - (B) NaHCO<sub>3</sub>
  - (C) Na<sub>2</sub>CO<sub>3</sub>.CaCO<sub>3</sub>
  - (D)  $Na_2CO_3$
- 55. Number of structural isomers of C<sub>4</sub>H<sub>10</sub> is
  - (A) 1
  - (B) 2
  - (C) 3
  - (D) 4

- 56. In an experiment, 1 ml of ethyl alcohol and 1 ml of acetic acid are taken in a test tube and 1 drop of concentrated sulphuric acid is added, ethylacetate is obtained. The process is known as
  - (A) Hydrolysis
  - (B) Esterification
  - (C) Saponification
  - (D) Neutralization
- 57. When vapours of ethyl alcohol are passed over heated alumina (Al<sub>2</sub>O<sub>3</sub>) above 350 °C, the compound which is obtained is
  - (A) acetaldehyde
  - (B) ethene
  - (C) formaldehyde
  - (D) ethane
- 58. Structural formula for propyne is
  - (A)  $CH_2 = C = CH_2$
  - (B)  $CH = CH = CH_2$
  - (C)  $CH_3 C \equiv CH$
  - (D)  $CH_3 \equiv C CH$

- 59. Which one of the following is a natural fibre?
  - (A) Cellulose
  - (B) Polythene
  - (C) Nylon
  - (D) Terylene
- 60. In natural gas, main constituent is
  - (A) butane
  - (B) iso-butane
  - (C) propene
  - (D) methane
- 61. Which of the following is bio-degradable?
  - (A) DDT
  - (B) Cotton
  - (C) Mercury
  - (D) Plastic

62.		ch of the following synthesizes ohydrates from inorganic raw	64.		ch of the following cell organelles have own DNA and ribosomes?
		rials?		(i)	Plastids
	(A)	Herbivores		(ii) ·	Golgi apparatus
	(B)	Carnivores		(iii)	Mitochondria
	(C)	Decomposers was all the second		(iv)	Endoplasmic reticulum
		Des Leaves to the later of the		(A)	(i) and (ii)
	(D)	Producers		(B)	(i) and (iii)
63.		ch groups of animals have a covering of es over the body?		(C)	(ii) and (iii)
				(D)	(ii) and (iv)
	(i)	Mammals (ii) Reptiles	65.	The	various aspects of scientific management
	(iii)	Amphibia (iv) Pisces			imal livestock, which include feeding, ding, housing and disease control are
	(A)	(i) and (ii)		term	ed as
	(B)	(ii) and (iii)		(A)	Cattle farming
		an a volumeland at damas (d)		(B)	Livestock farming
	(C)	(ii) and (iv)		(C)	Animal husbandry
	(D)	(i) and (iv)		(D)	Animal-based farming
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66.	Whic	ch variety of honeybee is commonly used	69.	The	structural and functional unit of kidney	
	for commercial honey production?			is		
	(A)	Apis florae		(A)	nephron	
		anner 1		(B)	neuron	
	(B)	Apis dorsata		(C)	Bowman's capsule	
	(C)	Apis mellifera		(D)	ureter	
	(D)	Apis cerana	70.	The breathing rate in aquatic organisms is more than terrestrial organisms because		
67.	The correct definition of vector is			(A)	the aquatic organisms require more	
	(A)	(A) infected person suffering from a communicable disease			oxygen to carry out their life process than terrestrial organism	
	(B)	microorganism which causes a disease in humans		(B)	lesser amount of oxygen remains dissolved in water than that present in air	
	(C)	an animal which transmits microbes from one person to another		(C)	the respiratory organs of aquatic organisms do not efficiently work	
	(D)	materials which cause allergies in		(D)	All of the above	
		humans	71.		ch component of food gets affected in	
68.	The human males after attaining puberty start				nouth cavity by the action of salivary lase?	
		ucing sperm in		(A)	Fats are broken down into fatty acids	
	(A)	epididymis		(77)	and glycerol	
	(B)	testes		(B)	Starch is broken down into sugars	
	(C)	prostrate gland		(C)	Proteins are broken down into amino acids	
	(D)	scrotal sac		(D)	Vitamins and minerals get absorbed	
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72.	Hormone that promotes growth in plants	74. A zygote which inherited 'X' chromosome
	is	from father will develop into a
	(A) T-RH	(A) girl
	(B) TSH	(B) boy
	(5)	(C) X-chromosome does not determine
	(C) abscisic acid	the sex of a child
	(D) auxin	(D) either boy or girl
		75. Oxygen is continuously being used by living
73.	The system of naming organisms by scientific	organisms but its percentage in the
names was introduced by	names was introduced by	atmosphere remains nearly the same. Which
		of the following processes is responsible for
	(A) Charles Darwin	the return of oxygen to the atmosphere?
	(B) Whittaker	(A) Respiration
	(C) Caralya Linnaya	(B) Fermentation
(	(C) Carolus Linneus	(C) Photosynthesis

(D) Robert Hooke

Burning of fossil fuels