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(Divisions of Horizon Study Circle Pvt. Ltd.)

Name.: Date : 25/10/2015

Test No.: 111

M.M.: 360

Subject Code.: 111

Time: 3 Hrs.

TES 7 for Engineering

Entrance Exam

2016

[Test No. 11]

INSTRUCTIONS FOR STUDENTS

- 1. Read each question carefully.
- 2. It is mandatory to use Blue/Black Ball Point Pen to darken the appropriate circle in the answer sheet.
- 3. Mark should be dark and should complete fill the circle.
- 4. Rough work must be done on the Question Paper, no additional sheet will be provided for this purpose.
- Do not use white-fluid or any other rubbing material on answer sheet. No change in the answer once marked.
- 6. Student cannot use log tables and calculators or any other material in the examination hall.

- 7. Before attempting the question paper, student should ensure that the test paper contains all pages and no page is missing.
- 8. Each correct answer carries four marks. One mark will be deducted for each incorrect answer from the total score.
- Before handing over the answer sheet to the invigilator, candidate should check the particulars have been filled and marked correctly.
- Immediately after the prescribed examination time is over, the answer sheet to be returned to the invigilator.
- 11. Use of Calculator and other Electronic device is not permitted.

Test No. 11

Topics of The Test

Physics	Work, Er	ergy and	Power + Cent				

Chemistry	Equilibrium.						
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Straight lines. Maths

Test No. 11

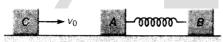
[PHYSICS]

- A ball is released from the top of a tower. The ratio of work done by force of gravity in first, second and third second of the motion of the ball is
 - (A) 1:2:3
- (B) 1:4:9
- (C) 1:3:5
- 1:5:3 (D)
- 2. Two springs have their force constants as k_1 and $k_2(k_1 > k_2)$, when they are stretched by the same force
 - (A) no work is done in case of both the springs
 - (B) equal work is done in case of both the springs
 - (C) more work is done in case of second spring
 - (D) more work is done in case of first spring
- A spring of force constant 800 Nm⁻¹ has an extension 3. of 5 cm. The work done in extending it from 5 cm to 15
 - (A) 16 J
- (B) 8 J
- (C) 32 J
- (D) 24 J
- A uniform chain of length 2 m is kept on a table such 4. that a length of 60 cm hangs freely from the edge of the table. The total mass of the chain is 4 kg. What is the work done in pulling the entire chain on the table?
 - (A) 7.2 J
- (B) 3.6 J
- (C) 120 J
- (D) 1200 J
- A particle is released from a height h. At a certain 5. height, its KE is two times its potential energy. Height and speed of the particle at that instant are

 - (A) $\frac{h}{3}$, $\sqrt{\frac{2gh}{3}}$ (B) $\frac{h}{3}$, $2\sqrt{\frac{gh}{3}}$
 - (C) $\frac{2h}{3}$, $\sqrt{\frac{2gh}{3}}$ (D) $\frac{h}{3}$, $\sqrt{2gh}$

- A ball of mass 2 kg and another of mass 4 kg are 6. dropped together from a 60 ft tall building. After a fall of 30 ft each towards earth, their respective kinetic energies will be in the ratio of
 - (A) $\sqrt{2}:1$
- (C) 1:2
- (D) $1:\sqrt{2}$
- 7. A particle is projected at 60° to the horizontal with a kinetic energy K. The kinetic energy at the highest point is
 - (A) K
- (B) zero
- (C) K/4
- (D) K/2
- 8. A vertical spring with force constant k is fixed on a table. A ball of mass m at a height h above the free upper end of the spring falls vertically on the spring, so that the spring is compressed by a distance d. The net work done in the process is
 - (A) $mg(h+d) + \frac{1}{2}kd^2$
 - (B) $mg(h+d)-\frac{1}{2}kd^2$
 - (C) $mg(h-d) \frac{1}{2}kd^2$
 - (D) $mg(h-d) + \frac{1}{2}kd^2$
- If F is the force required to keep a train moving at a 9. constant speed ν , the power required is
- (C) $\frac{1}{2}Fv$

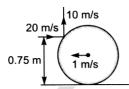
- 10. In two separate collisions, the coefficient of restitutions \mathbf{e}_1 and \mathbf{e}_2 are in the ratio 3:1. In the first collision the relative velocity of approach is twice the relative velocity of separation, then the ratio between relative velocity of approach and the relative velocity of separation in the second collision is
 - (A) 1:6
- (B) 2:3
- (C) 3:2
- (D) 6:1
- 11. A block C of mass m is moving with velocity v_0 and collides elastically with block A of mass m and connected to another block B of mass 2m through spring constant k. What is k if x_0 is compression of spring when velocity of A and B is same?



- $(A) \quad \frac{mv_0^2}{X_0^2}$
- (B) $\frac{mv_0^2}{2x_0^2}$
- (C) $\frac{3}{2} \frac{m v_0^2}{x_0^2}$
- (D) $\frac{2}{3} \frac{m v_0^2}{x_0^2}$
- 12. A bomb is kept stationary at a point. It suddenly explodes into two fragments of masses 1g and 3g. The total KE of the fragments is 6.4×10⁴ J. What is the KE of the smaller fragment?
 - (A) 2.5×10^4 J
- (B) $3.5 \times 10^4 \text{J}$
- (C) $4.8 \times 10^4 \text{J}$
- (D) 5.2×10⁴J
- 13. Consider elastic collision of a particle of mass m moving with a velocity u with another particle of the same mass at rest. After the collision the projectile and the struck particle move in directions making angles θ_1 and θ_2 respectively with the initial direction of motion. The sum of the angles $\theta_1 + \theta_2$ is
 - (A) 45°
- (B) 90°
- (C) 135°
- (D) 180°

- 14. A body of mass M moves with velocity v and collides elastically with another body of mass m(M >> m) at rest, then the velocity of body of mass m is
 - (A) v
- (B) 2v
- (C) v/2
- (D) zero
- 15. When U²³⁸ nucleus originally at rest, decays by emitting an alpha particle having a speed *u*, the recoil speed of the residual nucleus is
 - (A) $\frac{4u}{238}$
- (B) $-\frac{4u}{23^2}$
- (C) $\frac{4u}{234}$
- (D) $-\frac{4u}{238}$
- 16. An α -particle of mass m suffers one dimensional elastic collision with a nucleus of unknown mass. After the collision the α -particle is scattered directly backward losing 75% of its kinetic energy. Then the mass of the nucleus is
 - (A) m
- (B) 2 m
- (C) 3 m
- (D) $\frac{3}{2}m$
- 17. An elastic ball is dropped from a height *h* and it rebounds many times from the floor. If the coefficient of restitution is *e*, the time interval between the second and the third impact, is
 - (A) ev/g
- (B) $e^2 v / q$
- (C) $e^2 \sqrt{\frac{8h}{g}}$
- (D) $e^2 \sqrt{\left(\frac{h}{g}\right)^2}$

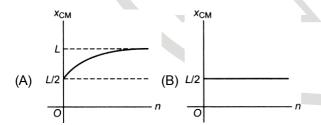
18. A thin ring of mass 2 kg and radius 0.5 m is rolling without slipping on a horizontal plane with velocity 1 m/s. A small ball of mass 0.1 kg, moving with velocity 20 m/s in the opposite direction, hits the ring at a height of 0.75 m and goes vertically up with velocity 10 m/s. Immediately after the collision

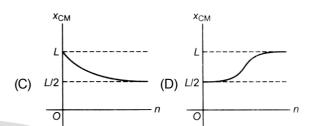


- (A) the ring has pure rotation about its stationary CM
- (B) the ring comes to a complete stop
- (C) the ring has pure translational motion
- (D) there is no friction between the ring and the ground
- 19. A system consists of 3 particles each of mass *m* located at points (1,1), (2,2) and (3,3). The coordinates of the centre of mass are
 - (A) (6,6)
- (B) (3,3)
- (C) (1,1)
- (D) (2,2)
- 20. A thin rod of length L is lying along the x-axis with its ends at x = 0 and x = L. Its linear density (mass\length)

varies with x as $k\left(\frac{x}{L}\right)^n$, where n can be zero or any

positive number. If the position $x_{\scriptscriptstyle CM}$ of the centre of mass of the rod is plotted against n, which of the following graphs best approximates the dependence of $x_{\scriptscriptstyle CM}$ on n?





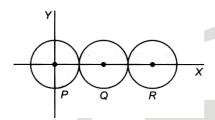
- 21. Two particles A and B initially at rest, move towards each other, under mutual force of attraction. At an instance when the speed of A is ν and speed of B is 2ν , the speed of centre of mass (CM) is
 - (A) zero
- (B) v
- (C) 2.5v
- (D) 4v
- 22. Four point masses *P*,*Q*,*R* and *S* with respective masses 1 kg, 1kg, 2kg and 2 kg situated on the corners of a square of side *a*. The centre of mass of the system will be farthest from
 - (A) Ponly
- (B) Rand S
- (C) Ronly
- (D) P and Q
- 23. Consider a two particle system with particles having masses m_1 and m_2 . If the first particles is pushed towards the centre of mass through a distance d, by what distance should the second particle be moved, so as to keep the centre of mass at the same position?
 - (A) $\frac{m_2}{m_1}$ d
- (B) $\frac{m_1}{m_1 + m_2}$
- (C) $\frac{m_1}{m_2}d$
- (D) d
- 24. A body A of mass M while falling vertically downwards under gravity breaks into two parts; a body B of mass

 $\frac{1}{3}M$ and a body C of mass $\frac{2}{3}M$. The centre of mass

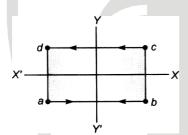
of bodies B and C taken together shifts compared to that of body A towards

- (A) depends on height of breaking
- (B) does not shift
- (C) body C
- (D) body B

25. Three identical spheres, each of mass 1 kg are kept as shown in figure below, touching each other, with their centres on a straight line. If their centres are marked P,Q,R respectively, the distance of centre of mass of the system from P is

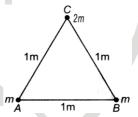


- 26. Four bodies of equal mass start moving with same speed are shown in the figure. In which of the following combination the centre of mass will remain at origin?



- (A) cd
- (B) ab
- (C) ac
- (D) bd
- 27. If linear mass density of a rod of length 3 m varies as $\lambda = 2 + x$, then the position of the centre of mass of the rod is

- 28. Two particles of equal mass have velocities $\mathbf{v}_1 = 4\hat{\mathbf{i}}$ and $\mathbf{v}_2 = 4\hat{\mathbf{j}} \text{ ms}^{-1}$. First particle has an acceleration $\mathbf{a}_1 = (5\hat{\mathbf{i}} + 5\hat{\mathbf{j}}) \,\text{ms}^{-2}$, while the acceleration of the other particle is zero. The centre of mass of the two particles moves in a path of
 - straight line
- (B) parabola
- (C) circle
- ellipse
- 29. The ratio of moments of inertia of circular ring and a circular disc having the same mass and radii about an axis passing through the centre and perpendicular to its plane is
 - (A) 1:1
- (B) 2:1
- 1:2
- (D) 4:1
- 30. Two balls each of mass m are placed on the vertices A and B of an equilateral triangle ABC of side 1m. A ball of mass 2m is placed at vertex C. The centre of mass of this system from vertex A (located at origin) is



- (C) $\left(\frac{1}{2}\text{m}, \frac{\sqrt{3}}{4}\text{m}\right)$ (D) $\left(\frac{\sqrt{3}}{4}\text{m}, \frac{\sqrt{3}}{4}\text{m}\right)$

[CHEMISTRY]

31. Consider the following gaseous equilibria with equilibrium constants K_1 and K_2 respectively.

$$SO_2(g) + \frac{1}{2}O_2(g) \rightleftharpoons SO_3(g)$$

$$2SO_3(g) \rightleftharpoons 2SO_2(g) + O_2(g)$$

The equilibrium constants are related as

$$(A) \quad 2K_1 = K_2$$

(A)
$$2K_1 = K_2^2$$
 (B) $K_1^2 = \frac{1}{K_2}$

(C)
$$K_2^2 = \frac{1}{K_1}$$
 (D) $K_2 = \frac{2}{K_1^2}$

(D)
$$K_2 = \frac{2}{K_1^2}$$

32. $2HI(g) \rightleftharpoons H_2(g) + I_2(g)$

> The equilibrium constant of the above reaction is 6.4 at 300 K. If 0.25 mole each of H₂ and I₃ are added to the system, the equilibrium constant will be

- (A) 6.4
- (B) 0.8
- (C) 3.2
- (D) 1.6
- The equilibrium constants K_{p_1} and K_{p_2} for the reactions 33.

 $X \rightleftharpoons 2Y$ and $Z \rightleftharpoons P + Q$, respectively are in the ratio of 1:9. If the degree of dissociation of X and Z be equal, then the ratio of total pressure at these equilibria is

- (A) 1:36
- (B) 1:1
- (C) 1:3
- (D) 1:9
- 34. The equilibrium constant (K_p) for the decomposition of gaseous H₂O

$$H_2O(g) \rightleftharpoons H_2(g) + \frac{1}{2}O_2(g)$$

is related to degree of dissociation (α) at a total pressure p is given by

(A)
$$K_{p} = \frac{\alpha^{3} p^{1/2}}{(1+\alpha)(2+\alpha)^{1/2}}$$

(B) $K_p = \frac{\alpha^3 p^{3/2}}{(1-\alpha)(2+\alpha)^{1/2}}$

(C)
$$K_{p} = \frac{\alpha^{3/2}p^{2}}{(1-\alpha)(2+\alpha)^{1/2}}$$

(D)
$$K_p = \frac{\alpha^{3/2} p^{1/2}}{(1-\alpha)(2+\alpha)^{1/2}}$$

- In the reaction, $3A + 2B \longrightarrow 2C$, the equilibrium constant K_c is given by

- 56 g of nitrogen and 8 g of hydrogen gas are heated in a closed vessel. At equilibrium 34 g of ammonia are present. The equilibrium number of moles of nitrogen, hydrogen and ammonia are respectively
 - (A) 1,2,2
- (B) 2,2,1
- (C) 1,1,2
- (D) 2,1,2
- 37. Partial pressure of O₂ in the reaction

$$2Ag_2O(s) \rightleftharpoons 4Ag(s) + O_2(g)$$
 is

- (C) $3\sqrt{K_0}$
- (D) 2K
- The equilibrium constant for the reaction, 38.

$$N_2(g) + O_2(g) \rightleftharpoons 2NO(g)$$

at temperature T is 4×10^{-4} . The value of K_c for the reaction

$$NO(g) \rightleftharpoons \frac{1}{2}N_2(g) + \frac{1}{2}O_2(g)$$

at the same temperature is

- (A) 2.5×10^2
- (B) 50
- (C) 4×10^{-4}
- (D) 0.02

- For the reaction, $C_2H_A + H_2 \rightleftharpoons C_2H_6$ the correct 39. relation is

 - (A) $K_{\rho} = K_{c}$ (B) $K_{\rho} = K_{c}[RT]$
 - (C) $K_p = K_c [RT]^{-2}$ (D) $K_p = K_c [RT]^{-1}$
- For which of the following reactions, $K_p = K_c$? 40.
 - (A) $N_2 + 3H_2 \rightleftharpoons 2NH_3$
 - (B) $N_2 + O_2 \rightleftharpoons 2NO$
 - (C) $PCI_5 \rightleftharpoons PCI_3 + CI_2$
 - (D) $2SO_3 \rightleftharpoons 2SO_2 + O_3$
- 41. Given the equilibrium system

$$NH_{A}CI(s) \rightleftharpoons NH_{A}^{+}(aq) + CI^{-}(aq)$$

$$(\Delta H = +3.5 \text{ kcal/mol}).$$

What change will shift the equilibrium to the right?

- (A) Decreasing the temperature
- (B) Increasing the temperature
- (C) Dissolving NaCl crystals in the equilibrium mixture
- (D) Dissolving NH₄NO₃ crystals in the equilibrium
- 42. In a lime kiln, to get higher yield of CO₂, the measure that can be taken is
 - (A) to remove CaO
 - (B) to add more CaCO₂
 - (C) to maintain high temperature
 - (D) to pump out CO₂
- 43. What is the best description of the change that occurs when Na₂O(s) is dissolved in water?
 - (A) Oxidation number of sodium decreases
 - (B) Oxide ion accepts sharing in a pair of electrons
 - (C) Oxide ion donates a pair of electrons
 - (D) Oxidation number of oxygen increases

44. The strongest base is



- 45. Four species are listed below
 - HCO,
- H,O
- (iii) HSO,
- HSO,F

Which one of the following is the correct sequence of their acid strength?

- (A) (iv) < (ii) < (iii) < (i)
- (B) (ii) < (iii) < (i) < (iv)
- (C) (i) < (iii) < (ii) < (iv)
- (D) (iii) < (i) < (iv) < (ii)
- 46. The first and second dissociation constants of an acid $H_{\circ}A$ are 1.0×10^{-5} and 5.0×10^{-10} respectively. The overall dissociation constant of the acid will be
 - (A) 5.0×10⁻⁵
- (B) 5.0×10^{15}
- (C) 5.0×10^{-15}
- (D) 0.2×10⁵
- 47. Arrange NH₄,H₂O,H₃O+,HF and OH- in increasing order of acidic nature
 - (A) $H_3O^+ < NH_4^+ < HF < OH^- < H_2O$
 - (B) $NH_4^+ < HF < H_3O^+ < H_2O < OH^-$
 - (C) $OH^- < H_2O < NH_4^+ < HF < H_3O^+$
 - (D) $H_2O^+ > HF > H_2O > NH_4^+ > OH_2^-$
- 48. The species among the following, which can act as an acid and a base is
 - (A) HSO
- (B) SO₄²
- (C) H₃O+

- 49. The values of dissociation constant of bases are given below. Which is the weakest base?
 - (A) 1.8×10⁻⁵
- (B) 4.8×10^{-10}
- (C) 7.2×10⁻¹¹
- (D) 7.07×10⁻⁷
- 50. The strongest conjugate base is
 - (A) NO₃
- (B) CI
- (C) SO₄²
- (D) CH₃COO
- 51. The pH of 10⁻¹⁰ M NaOH solution is nearest to
 - (A) -4
- (B) -10
- (C) 4
- (D) 7
- 52. 10⁻⁶ M NaOH is diluted 100 times. The pH of the diluted base is
 - (A) between 7 and 8 (B) between 5 and 6
 - (C) between 6 and 7 (D) between 10 and 11
- 53. 100 mL of 0.015 M HCl solution is mixed with 100 mL of 0.005 M HCl. What is the pH of the resultant solution?
 - (A) 2.5
- (B) 1.5
- (C) 2
- (D) 1
- 54. At 25° C, the solubility product of $Mg(OH)_2$ is 1.0×10^{-11} . At which pH, will Mg^2 ions start precipitating in the form of $Mg(OH)_2$ from a solution of $0.001 \, M \, Mg^2$ ions?
 - (A) 9
- (B) 10
- (C) 11
- (D) 8
- 55. Approximate relationship between dissociation constant of water (K_{ω}) and ionic product of water (K_{ω}) is
 - (A) $K_w = K$
- (B) $K_{w} = 55.6 \times K$
- (C) $K_w = 18 \times K$
- (D) $K_{w} = 14 \times K$

56. The pH of an aqueous solution of CH_3COONa of concentration C(M) is given by

(A)
$$7-\frac{1}{2}pK_a-\frac{1}{2}\log C$$

(B)
$$\frac{1}{2}pK_{w} + \frac{1}{2}pK_{b} + \frac{1}{2}\log C$$

(C)
$$\frac{1}{2}pK_{w} - \frac{1}{2}pK_{b} - \frac{1}{2}\log C$$

(D)
$$\frac{1}{2}pK_w + \frac{1}{2}pK_a + \frac{1}{2}\log C$$

- 57. The pK_a of a weak acid, HA, is 4.80. The pK_b of a weak base, BOH is 4.78. The pH of an aqueous solution of the corresponding salt, BA, will be
 - (A) 9.58
- (B) 4.79
- (C) 7.01
- (D) 9.22
- 58. The hydrolysis of sodium carbonate involves the reaction between
 - (A) sodium ion and water
 - (B) Na⁺ and OH[−]
 - (C) CO₃²⁻ and water
 - (D) CO₃²⁻ and H⁺
- 59. 0.1 mole of $CH_3NH_2(K_b = 5 \times 10^{-4})$ is mixed with 0.08 mole of HCl and diluted to 1 L. What will be the H⁺ concentration in the solution?
 - (A) $8 \times 10^{-2} \,\mathrm{M}$
- (B) $8 \times 10^{-11} \text{ M}$
- (C) $1.6 \times 10^{-11} \,\mathrm{M}$
- (D) $8 \times 10^{-5} \text{ M}$
- 60. Degree of dissociation of NH_4OH in water is 1.8×10^{-5} , then hydrolysis constant of NH_4CI is
 - (A) 1.8×10⁻⁵
- (B) 1.8×10⁻¹⁰
- (C) 5.55×10⁻⁵
- (D) 5.55×10⁻¹⁰

[MATHEMATICS]

- 61. The straight line 3x + y = 9 divides the line segment joining the points (1,3) and (2,7) in the ratio
 - (A) 3:4 externally
- (B) 3:4 internally
- (C) 4:5 internally
- (D) 5:6 externally
- 62. If the slope of one of the lines represented by $ax^2 + 2hxy + by^2 = 0$ be the square of the other, then

$$\frac{a+b}{h} + \frac{8h^2}{ab}$$
 is

- (A) 3
- (B) 4
- (C) 5
- (D) 6
- If the lines kx-2y-1=0 and 6x-4y-m=0 are 63. identical (coincident) lines, then the values of k and m
 - (A) k = 3, m = 2
- (B) k = -3, m = 2
- (C) k = -3, m = -2 (D) k = 3, m = -2
- 64. The perpendicular bisector of the line segment joining P(1,4) and Q(k,3) has y-intercept -4. Then, a possible value of k is
 - (A) 4
- (B) 1
- (C) 2
- (D) -2
- The value of k such that the lines 2x 3y + k = 0, 65. 3x - 4y - 13 = 08x - 11y - 33 = 0and are concurrent, is
 - (A) 20
- (B) -7
- (C) 7
- (D) -20
- 66. The equation of perpendicular bisectors of sides AB and AC of a $\triangle ABC$ are x-y+5=0 and x+2y=0respectively. If the coordinates of vertex A are (1,-2), then equation of BC is
 - (A) 14x + 23y 40 = 0
 - (B) 14x 23y + 40 = 0
 - 23x + 14y 40 = 0(C)
 - (D) 23x 14y + 40 = 0

- 67. The line passing through the point of intersection of x + y = 2, x - y = 0 and is parallel to x + 2y = 5, is
 - (A) x + 2y = 1
- (B) x + 2y = 2
- (C) x + 2y = 4
- (D) x + 2y = 3
- 68. If the lines $3x + 4y + 1 = 0.5x + \lambda y + 3 = 0$ and 2x + y - 1 = 0 are concurrent, then λ is equal to
 - (A) 8
- (B) 8
- (C) 4
- (D) -4
- A straight line through the point A(3,4) is such that its 69. intercept between the axes is bisected at A. Its equation is
 - (A) 3x-4y+7=0 (B) 4x+3y=24
- - (C) 3x + 4y = 25
- (D) x + v = 7
- The equation of the straight line passing through the point (4,3) and making intercepts on the coordinate axes whose sum is -1, is
 - (A) $\frac{x}{2} + \frac{y}{3} = -1$ and $\frac{x}{-2} + \frac{y}{1} = -1$
 - (B) $\frac{x}{2} \frac{y}{3} = -1$ and $\frac{x}{-2} + \frac{y}{1} = -1$
 - (C) $\frac{x}{2} + \frac{y}{3} = 1$ and $\frac{x}{-2} + \frac{y}{1} = 1$
 - (D) $\frac{x}{2} \frac{y}{3} = 1$ and $\frac{x}{2} + \frac{y}{1} = 1$
- 71. The equation of pair of lines joining origin to the points of intersection of $x^2 + y^2 = 9$ and x + y = 3 is
 - (A) $x^2 + (3 x)^2 = 9$ (B) xy = 0
 - (C) $(3+y)^2 + y^2 = 9$ (D) $(x-y)^2 = 9$

- 72. The equation of the pair of straight lines perpendicular to the pair $2x^2 + 3xy + 2y^2 + 10x + 5y = 0$ and passing through the origin, is
 - (A) $2x^2 + 5xy + 2y^2 = 0$
 - (B) $2x^2 3xy + 2y^2 = 0$
 - (C) $2x^2 + 3xy + y^2 = 0$
 - (D) $2x^2 5xy + 2y^2 = 0$
- If the equation $4x^2 + hxy + y^2 = 0$ represent coincident 73. lines, then h is equal to
 - (A) 1
- (B) 3
- (C) 2
- (D) 4
- 74. A straight line through P(1,2) is such that its intercept between the axes is bisected at P. Its equation is
 - (A) x + y = -1
- (B) x+y=3
- (C) x + 2y = 5
- (D) 2x + y = 4
- If the equation $kx^2 2xy y^2 2x + 2y = 0$ represents 75. a pair of lines, then k is equal to
 - (A) 2
- (B) -2
- (C) -5
- (D) 3
- The line L given by $\frac{x}{5} + \frac{y}{h} = 1$ passes through the point (13,32). The line K is parallel to L and has the equation
 - $\frac{x}{2} + \frac{y}{2} = 1$. Then, the distance between L and K is

- 77. The number of points on the line x + y = 4 which are unit distance apart from the line 2x + 2y = 5 is
 - (A) 0
- (B) 1
- (C) 2
- (D) ∞
- The coordinates of the foot of perpendicular from (a,0) 78. on the line $y = mx + \frac{a}{m}$ are
 - - $\left(0,\frac{a}{m}\right)$ (B) $\left(0,-\frac{a}{m}\right)$
- 79. The distance between the lines 3x + 4y = 9 and 6x + 8y = 15 is
- (C) 6
- (D) None of these
- 80. If A(2,-1) and B(6,5) are two points, then the ratio in which the foot of the perpendicular from (4,1) to AB divides it, is
 - (A) 8:15
- (B) 5:8
- (C) -5:8
- (D) -8:5
- 81. The distance between the pair of parallel lines $x^{2} + 2xy + y^{2} - 8ax - 8ay - 9a^{2} = 0$ is
 - (A) $2\sqrt{5}a$
- (B) $\sqrt{10}a$
- (C) 10a
- (D) $5\sqrt{2}a$
- The distance between the pair of parallel lines given 82. by $x^2 - 1005x + 2006 = 0$ is
 - (A) 1001
- (B) 1000
- (C) 1005
- (D) 2006

- 83. If the equation of base of an equilateral triangle is 2x - y = 1 and the vertex is (-1,2), then the length of the side of the triangle is

- 84. The product of the perpendicular distances from the origin on the pair of straight $12x^2 + 25xy + 12y^2 + 10x + 11y + 2 = 0$ is

- 85. Distance between the two parallel lines y = 2x + 7 and y = 2x + 5 is
 - $\sqrt{5}/2$
- (C) $2/\sqrt{5}$
- (D) 1/√5
- 86. All chords of the curve $3x^2 - y^2 - 2x + 4y = 0$ which subtend a right angle at the origin, pass through the fix point
 - (A) (1,2)
- (B) (1,–2)
- (C) (-1,2)
- (D) (-1,-2)

- If the lines $px^2 qxy y^2 = 0$ make the angle α and 87. β with x-axis, then the value of $tan(\alpha + \beta)$ is

- (D) $\frac{-p}{1+q}$
- A pair of perpendicular straight lines passes through 88. the origin and also through the point of intersection of the curve $x^2 + y^2 = 4$ with x + y = a. The set containing the value of 'a' is
 - $(A) \{-2,2\}$
- (B) $\{-3,3\}$
- (C) {-4,4}
- (D) {-5,5}
- 89. The bisector of the acute angle formed between the lines 4x-3y+7=0 and 3x-4y+14=0 has the equation
 - (A) x + y + 3 = 0
- (B) x y 3 = 0
- (C) x y + 3 = 0
- (D) 3x + y 7 = 0
- 90. The equation of the line bisecting perpendicularly the segment joining the points (-4,6) and (8,8) is
 - (A) y = 7
- (B) 6x + y 19 = 0
- (C) x+2y-7=0 (D) 6x+2y-19=0