SAMPLE PAPER

Time : 3 :00 hours. (Class 12th) Maximum Marks : 360

INSTRUCTIONS

- 1. Immediately fill in the particulars on this page of the Test Booklet with **Black Ball Point Pen**. Use of pencil is strictly prohibited.
- 2. The Answer Sheet is kept inside this Test Booklet. When you are directed to open the Test Booklet, take out the Answer Sheet and fill in the particular carefully.
- 3. The test is of 3 hours duration.

NOT BREAK THE SEAL WITHOUT BEING INSTRUCTED TO DO SO BY THE INVALIDATOR

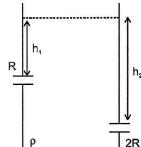
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- 4. The Test Booklet consists of 90 questions. The maximum marks are 360.
- 5. There are three parts in the question paper A, B, C consisting of **Physics, Chemistry** and **Mathematics** having total 30 questions in each part of equal weightage. Each question is allotted **4 (four) marks** for correct response.
- 6. Candidates will be awarded marks as stated above in Instructions No. 5 for correct response of each question. ¼ (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
- 7. **There is only one correct response for each question.** Filling up more than one response in any question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instructions 6 above.
- 8. No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc., except the Admit Card inside the examination room/hall.
- Rough work is to be done on the space provided for this purpose in the Test Booklet only. This space is given at the bottom of each page and in one page at the end of the booklet.
- 10. On completion of the test, the candidate must hand over the Answer Sheet to the Invigilator on duty in the Room/ Hall. *However, the candidates are allowed to take away this Test Booklet with them.*
- 11. The CODE for this Booklet **0**. Make sure that the CODE printed on the Answer Sheet is the same as that on this booklet. In case of discrepancy, the candidate should immediately report the matter to the Invigilator for replacement of both the Test Booklet and the Answer Sheet.
- 12. Do not fold or make any stray marks on the Answer Sheet.

Name of the Candiate (in Capital letters) :								
Roll Number : in figures : in words :								
Name of Examination Centre (in Capital letters) :								
Candidate's Signature : Invigilator's Signature :								

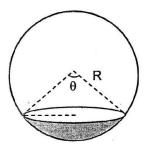
- Q.1 An explosion breaks a rock into three parts. Two parts of masses 1 kg and 2 kg go off at right angles to each other with velocity of 12 m/s and velocity of 8 m/s respectively. If the third part flies off with a velocity of 4 m/s, its mass would be:
 - (A) 7 kg

- (B) 17 kg
- (C) 3 kg
- (D) 5 kg
- Q.2 In a cylindrical water tank there are two small holes of radius R and 2R respectively at a depth $\,h_1$ and $\,h_2$ from the upper level of water. If tank is at rest on smooth horizontal surface than the ratio of $\,h_1$ and $\,h_2$: (Assume that height of liquid level is kept constant) R



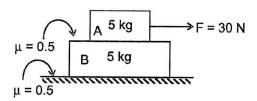
- (A)
- (B) 2
- (C) 3
- (D) 4
- Q.3 Assume that magnetic field of earth is similar to magnetic field of a magnetic bar. If angle of dip at Delhi and Chennai are θ_1 are θ_2 respectively then:
 - (A) $\theta_1 > \theta_2$
- (B) $\theta_1 = \theta_2$
- (C) $\theta_1 < \theta_2$
- (D) $2\theta_1 = \theta_2$
- Q.4 A particle is projected with initial speed 10 m/s at an angle of 53° with horizontal. Then the time after which its speed becomes half of the initial speed.
 - (A) 1 sec
- (B) $\frac{4}{5}$ sec
- (C) $\frac{8}{5}$ sec
- (D) Never

Q.5 Consider a spherical drop of radius R. Surface tension of the liquid is S. Force of surface tension on shaded sub hemisphere due to remaining drop is $\frac{S\pi R}{2}$. The value of angle $_{\theta}$ (angle subtended by the sub hemisphere at centre of drop) is :



(A) 60°

- (B) 120°
- (C) 30°
- (D) 90°
- Q.6 Consider a system of two blocks as shown in the figure. The accelerations of both the blocks (A and B) respectively are



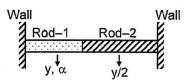
(A) 0;0

(B) $0:1 \text{ m/s}^2$

(C) 1 m/s^2 ; 0

(D) 1;1

- Q.7 If on increasing temperature of the systme shown, interface does not displace, thermal coefficient of linear expansion of rod-2 is



(A) α (B) 2α

- (C)

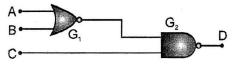
8 kg

- Q.8 In the system shown, the mass m = 2kg oscillates in a circular arc of amplitude 60°, the minimum value of coefficient of friction between mass = 8 kg and surface of table to avoid slipping is:
 - (A) 0.25

(B) 0.50

(C) 0.40

- (D) 0.30
- Q.9 For the given combination of gates, if the logic states of inputs A, B, C are as follows A = B = C = 0 and A = B = 1, C = 0 then the logic states of output D are



(A) 0, 0

(B) 0, 1

- (C) 1, 0
- (D) 1, 1

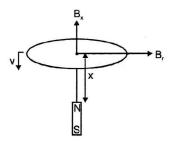
2 kg

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Q.10	If the mass of proton = 1.00	08 a.m.u. a	and mass of neut	tron = 1.009	a.m.u. then bindir	ng energy _l	per nucleon for	– ₄ Be ⁹
	(mass = 9.012 amu) would	ld be						
	(A) 0.065 MeV	(B)	60.44 MeV	(C)	67.2 MeV	(D)	6.72 MeV	
Q.11	When light of intensity 1 W	//m² and	wavelength 5 ×	10 ⁻⁷ m is inc	cident on a surfac	ce, it is con	npletely absorb	ed by
	the surface. If 100 photons	s emit one	electron and are	ea of the sur	face is 1 cm ² , the	en the pho	toelectric curre	nt will
	be							
	(A) 2 mA	(B)	0.4 μΑ	(C)	4.0 mA	(D)	4 μΑ	
Q.12	If the frequency of K_{α} line	of the eler	nent having Z =	65 is υ_{α} , an	d the frequency o	of K_{α} line of	of the element h	aving
	$Z=81$ is υ_{α} , then the rati	o of their	frequencies will	be about				
	(A) $\sqrt{2/5}$	(B)	2/√5	(C)	16/25	(D)	2/5	
Q.13	When 100 volt d.c. is app	lied acros	ss solenoid a cu	rrent of 1.0 a	amp flows in it. V	When 100	volts a.c. is ap	plied
	across the same coil, the current drops to 0.5 amp. If the frequency of the a.c. source is 50 Hz the impendance							
	and inductance of the sole	noid are						
	(A) 200 ohm and 0.55 He	enry		(B)	100 ohm and 0.	86 Henry		
	(C) 200 ohm and 1.0 Her	nry		(D)	100 ohm and 0.	93 Henry		
Q.14	A particle of mass 1.6 × 1	10 ⁻²⁷ kg a	ind charge 1.6	× 10 ⁻¹⁹ coul	omb enters a ur	niform 🔽	N × ×	× ×
	magnetic field of 1 Tesla as shown in the figure. The speed of the particle is 10^7m/s . the							
	distance PQ will be						$\stackrel{Q}{\times} \stackrel{\widehat{\times}}{\times}_{B}$	x x x x
	(A) 0.14 m			(B)	0.28 m			× ×

(D) 0.5 m

(C) 0.4 m

A conducting ring of radius r and resistance R is moving downward with a constant velocity v. Let due to a fixed Q.15 magnet axial magnetic field varies only with the axial distance (x) from north pole as $B_x = \frac{1}{\pi x}$ Tesla and the radial magnetic field variws as $B_r = \frac{2}{\pi x}$ Tesla. When the ring is at a height x from the north pole of the magnet, the magnetic force experienced by the ring is:



(A) $\frac{2r^3v}{Rv^3}$ upwards

(B) $\frac{2r^3v}{Rv^3}$ downwards

(C) $\frac{4r^3v}{Rv^3}$ upwards

- (D) $\frac{4r^3v}{Rv^3}$ downwards
- The radius of the earth is $\,{\rm R}_{\rm e}\,$ and the acceleration due to gravity at its surface is g. The work required in raising a Q.16 body of mass m to a height h from the surface of the earth will be

- (B) $\frac{\text{mgh}}{\left(1+\frac{h}{R_e}\right)^2}$ (C) $\frac{\text{mgh}}{\left(1+\frac{h}{R_e}\right)}$ (D) $\frac{\text{mg}}{\left(1+\frac{h}{R_e}\right)}$

258 Hz

1%

(D)

Q.19 An unknown quantity x is measured using an experiment by measuring a length ℓ (in cm) from scale having least count of 1 cm. Formula used is $x = R \frac{\ell}{100 - \ell} \cdot R$ is known accurately. Find the percentage error in measurement of 'x' for $\ell = 50$ cm.

(C) 242 Hz

(B) 2 % (C) 3 % (A) 4 %

(B) 262 Hz

Q.20 Light traveling in air falls at an incidence angle of 2° on one refracting surface of a prism of refractive index 1.5 and angle of refraction 4° . The medium on the other side is water (n = 4/3). Find the deviation produced by the prism.

(A) 1 (B) 2 (C) 3 (D) 4

Q.21 A mono atomic ideal gas is filled in a non conducting container. The gas can be compressed by a movable non conducting piston. The gas is compressed slowly to 12.5% of its initial volume, the ratio of initial adiabatic bulk modulus of the gas to the final value of adiabatic bulk modulus of the gas is:

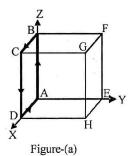
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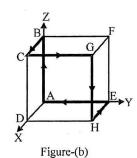
(A)

282 Hz

- (A) the width of the central maxima becomes double
- (B) the widht of the central maxima becomes half
- (C) the width of the central maxima becomes one fourth
- (D) the with of the cental maxima becomes four times.

Q.23 Current I is flowing along the path ABCD, along the four edges of the cube (figure-a), produces a magnetic field B₀ at the centre of the cube. Find the magnetic field B produced at the center of the cube by a current I flowing along the path of the six edges ABCGHEA (figure-b)



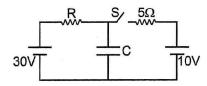


- (A) $\sqrt{\frac{3}{2}} B_0$ directed towards corner G
- (B) $\sqrt{3} B_0$ directed towards corner E

(C) $\sqrt{\frac{3}{2}}B_0$ directed towards corner H

(D) $\sqrt{3} B_0$ directed towards corner F

Q.24 In the circuit shown the capacitor is in steady state. The current through the 5Ω resistance just after the switch S is closed is:



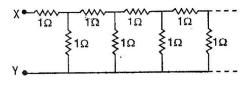
(A) 2 A

(B) 4 A

(C) depends on value of R only

- (D) depends on the value of R and C
- Q.25 An electric field given by $\vec{E} = 4 \ \hat{i} 3(y^2 + 2) \ \hat{j}$ passes Gaussian cube of side 1 m placed with one corner at origin such that its sides represents x, y and z axes. The magnitude of net charge enclosed within the cube is
 - (A) $4 \varepsilon_0$

- (B) $3\varepsilon_0$
- (C) $5\varepsilon_0$
- (D) $-2\varepsilon_0$
- Q.26 Figure shows an infinite ladder network of resistances. The equivalent resistance between points X and Y is



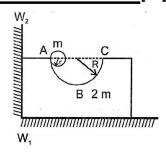
(A) Infinite

(B) 3 ohm

(C) 8.62 ohm

(D) 1.62 ohm

Q.27 A sphere of mass m and radius r is released from a wedge of mass 2m as shown. ABC is hemispherical positon of radius R. The impulse imparted to the system consisting wedge and sphere by the vertical wall w₁ w₂ till the time sphere reaches at the bottom most position of spherical prtion for the first time is: (Friction between wedge and horizontal surface is absent and between sphere and wedge friction is sufficient to avoid slipping between them.)



- $\text{(A)} \quad m \sqrt{\frac{2g(R-r)}{7}} \\ \text{(B)} \quad m \sqrt{\frac{10g(R-r)}{7}} \\ \text{(C)} \quad 2m \sqrt{\frac{5g(R-r)}{7}} \\ \text{(D)} \quad 3m \sqrt{\frac{2g(R-r)}{7}} \\$
- Q.28 Angular magnification produced by astronomical telescope for normal adjustment is 10 and length of telescope is 1.1 m. The angular magnification when the image is formed at least distance of distinct vision (D = 25 cm) is
 - (A) 12

- (C) 16
- Q.29 Two S.H.Ms. $s_1 = a \sin \omega t \& s_2 = b \sin \omega t$ are superimposed on a particle. The s_1 and s_2 are along the directions which make angle 37° with each other.
 - (A) the particle will perform S.H.M.

- (B) the particle will not perform S.H.M.
- the particle will perform motion but not S.H.M.
- (D) the motion will not be oscillatory.
- Q.30 A T.V tower has a height of 100 m. The population which is covered by T.V. broadcast, if the average population density around the tower is 1000/km² will be
 - (A) 39.5×10^5
- (B) 40.2×10^5
- (C) 29.5×10^7

PART - II CHEMISTRY

- Q.31 A sample of 3 gram (92% pure) of silver carbonate on being strongly heated yields a residue and gaseous products Find the volume of evolved gases at STP.
 - (A) 112 mL
- (B) 22 mL
- (C) 336 mL
- (D) 33.6 mL
- Q.32 The enthalpy of combustion of cyclohexane, 1,4-cyclohexadiene and H_2 are respectively -3920, -3680 and -241 KJ mol⁻¹. The heat of hydrogenation of 1, 4-cyclohexadiene is:
 - (A) -121 KJ/mol
- (B) -242 KJ/mol
- (C) 42 KJ/mol
- (D) 242 KJ/mol
- Find the freezing point of a solution when 5 mol of $K_{\rm I}$ is mixed with 1 mol of $HgCl_2$ in 3500 gram of H_2O . $K_{\rm f}$ of Q.33

$$H_2O = 1.86 \frac{K}{m}$$
. (KI + HgCl₂ \longrightarrow K₂Hgl₄ + KCl)

- (A) -28.7° C
- (B) −41.45°C
- (C) -35.07° C
- (D) -4.78° C
- For the following gaseous equilibrium w, x, y and z at 300 K, the increasing order of $\left(\frac{K_P}{K_C}\right)$ is : Q.34

$$W: 2HI \square \square \square \square \square \square \square \square \square$$

$$x: O_2 + 2SO_2 \oplus \oplus \oplus 2SO_3$$

$$y : PCl_5 \square \square \square PCl_3 + Cl_2$$

$$z: N_2 + 3H_2 \square \square \square$$
 2NH₃

$$(A) \quad w = x = y = z$$

(B)
$$z < x < w < y$$

(C)
$$y < w < x < z$$
 (D) $z < x < w = y$

(D)
$$z < x < w = v$$

- Q.35 The slag produced during formation of wrought iron from its ore is/are:
 - (A) CaSiO₃ & MnAsO₃

(B) CaSiO₃

(C) MnSiO₃

(D) CaSiO₃ & MnSiO₃

- Q.36 Bleaching powder is a mixture of :
 - (A) Ca(OCI)₂

(B) $Ca(OCI)_2 + Ca(OH)_2$

(C) $CaCl_2 + 2H_2O$

- (D) $Ca(OCI)_2 + CaCI_2 + Ca(OH)_2 + 2H_2O$
- Q.37 Given $E^o_{Cu^+/Cu} = 0.52\,V, \; E^o_{Cu^{2_+}/Cu^+} = 0.16\,\,V$. The equilibrium constant for disproportionation reaction

$$2Cu^{+}(aq) \longrightarrow Cu(s) + Cu^{2+}(aq)$$
 at 298 K is : (Take $\frac{2.303RT}{F} = 0.06 \text{ V}$)

- (A) 6×10^4
- (B) 6×10^6
- (C) 10^6
- (D) 1.2×10^{-6}

- Q.38 Correct statement about helmoholtz electrical double layer is :
 - (I) It is a combination of two layer of similar charges around colloidal sol.
 - (II) It is a combination of two layers of opposite charges around the colloidal sol.
 - (III) In it 1st layer of ions is diffused while 2nd layer of ions is fixed.
 - (IV) The potential difference between the fixed layer and the diffused layer is called zeta potential.
 - (A) II & IV
- (B) I & III
- (C) III & IV
- (D) I & IV

- Metal carbonyls possess both σ and π character.
- Synergic bonding is present in metal carbonyls.
- $M-C\pi$ bond in metal carbonyls is formed by the donation of pair of e⁻ from a filled d-orbital of metal to vacant antibonding orbital (π) in CO.
- (D) All of these

In FCC structure what is the total number of voids in 0.6 mol of compound ? $\{N_A = 6 \times 10^{23} \text{ / mol}\}$ Q.40

- (A) 6.00×10^{23}
- (B) 1.08×10^{24}
- (C) 1.08×10^{23} (D) 7.2×10^{24}

If the e^- is transferred from A to B forming A^+ & B^- . The above process is possible when: Q.41

[I = Ionisation enthalpy, E = Magnitude of electron gain enthalpy]

(A) $(E_B + I_B) = (I_A - E_A)$

(B) $(I_A - E_B) > (I_B - E_A)$

(C) $(I_B + E_B) > (I_A + E_A)$

(D) None of these

Q.42 In a cyclotimetaphosphoric acid molecule 3 double bonds; 12 single bonds are present.

- PH₃ is highly poisonous and has smell like rotten fish.
- Strong reducing behaviour of H_3PO_2 is due to presence of two P-H bonds. (III)
- (IV) SO₂ acts as bleaching agent in moist conditions.
- SO_2 can be prepared by the reaction of dilute H_2SO_4 with metal sulphide.

Correct statements are:

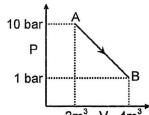
- (A) I, II & III
- (B) I, II, III & IV
- (C) I, II, III, IV & V
- (D) II. III & V

- Q.43 In the decomposition of Ammonia it was found that at 50 torr pressure $t_{1/2}$ was 3.64 hour while at 100 torr $t_{1/2}$ was 1.82 hours. Then order of reaction would be :
 - (A) 0

(B) 1

- (C) 2
- (D) 3

- Q.44 Which of the following statement is INCORRECT?
 - (A) $\rm C_{60}\,$ contains twenty-six membered rings and twelve-five membered rings.
 - (B) Fullerenes are cage-like molecules.
 - (C) Graphite is thermodynmically most stable allotrope of carbon.
 - (D) None opf these
- Q.45 A buffer solution cannot be prepared by mixing solutions of :
 - (A) Sodium chloride and Sodium hydroxide
 - (B) Ammonium hydroxide and Ammonium chloride.
 - (C) Formic acid and Sodium format.
 - (D) Boric acid and borax.
- Q.46 What is the value of ΔH_{AB} (in KJ) for an ideal gas in the given process represented in the figure ? Heat supplied during the process = 2740 KJ



(A) 40 KJ

(B) 400 KJ

(C) 1640 KJ

(D) 80 KJ

- Q.47 Gadolinium (Gd) belongs to 4f series. It's atomic number is 64. Which of the following is the correct electronic configuration of gadolinum?
 - (A) [Xe] $4f^7$, $5d^1$, $6s^2$

(B) [Xe] $4f^6$, $5d^2$, $6s^2$

(C) [Xe] $4f^8$, $6d^2$

- (D) [Xe] 4f⁹, 5s¹
- Q.48 In which of the following molecule/ion no unpaired electron is present?
 - (A) N_2^+

(B) O₂

- (C) O_2^{2-}
- (D) B₂
- Q.49 Toluene is nitrated and the resulting product is reduced with tin and hydrochloric acid. The product so obtained is diazotised and then heated with cuprous bromide. The reaction mixture so formed contains:
 - (A) mixture of o- and p-dibromobenzenes
- (B) mixuture of o-and p-bromoanilines

(C) mixture of o-and m-bromotoluenes

- (D) mixture of o-and p-bromotoluenes
- Q.50 The pyrimidine bases present in DNA are
 - (A) cytosine and guanine

(B) cytosine and thymine

(C) cytosine and uracil

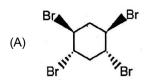
(D) cytosine and adenine

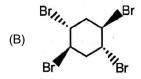
- Q.51 Choose the correct option:
 - (A) Teflon, Polystyrene and neoprene are Copolymers

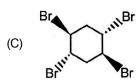
(B) Monomer of
$$\begin{bmatrix} CH_3 \\ CH_2 - C \\ CH_3 \end{bmatrix}_n$$
 is butane

- (C) Bakelite is a cross linked polymer
- (D) Olefinic monomers undergo condensation polymerisation

Q.52 → Which product is not possible in above reaction?

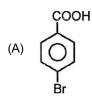


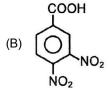




- ,ii Br (D) N_{Br}
- Q.53 Identify the molecule not showing resonance.
 - (A) $CH_2 = C = CH_2$
- (B) $CH_2 = C = O$
- $CH_2 = \overset{\oplus}{N} = \overset{\Theta}{N}$ (C)
- (D) O = C = O

Q.54 Which one is the strongest acid.







- COOH (D)
- Q.55 Trichloroacetaldehyde was subjected to reaction by using NaOH. The mixture of the products contains sodium trichloroacetate ion and another compound. The other compound is:
 - 2, 2, 2 Trichloroethanol

(B) Trichloromethanol

(C) 2, 2, 2-Trichloropropanol

(D) Chloroform

Q.56
$$CH_3$$
 $C=C$ CH_3 X $C=C$ CH_3 $C=C$ CH_3 $C=C$ $C-ONa$ CH_3 $C=C$ $C-ONa$

Reagent X is:

- (A) O_3 / Zn, H_2O
- (B) NaOX
- (C) CrO_3 / CH_2CI_2
- (D) Cu/Δ
- Q.57 What is the end product if alcohol consist primary alkyl group and react with H₂SO₄ at low temperature (413 K):
 - (A) Alkene

(B) Protonated alcohol

Ether (C)

- (D) Alkyl hydrogen sulphate
- Q.58 When Me₃C - O - CH₃ react with excess of HI gives CH₃I and Me₃CI. It follows the type of mechanism respectively
 - (A) only $S_N 1$
- (B) only $S_N 2$
- (C) $S_N = 2 \& S_N = 1$ (D) $S_N = 1 \& S_N = 2$
- Q.59 Which is the correct code for the given name reactions?
 - $PH = C \equiv N \frac{\text{(i) SnCl}_2 / HCl}{\text{(ii) H}_2O}$

Perkin reaction

 $Ph-CHO \xrightarrow{KCN/\Delta}$ (q)

(ii) Benzoin condensation

 $Ph-CHO \xrightarrow{NaOH}$

(iii) Stephen reduction

Ph – CHO $\xrightarrow{\text{(CH}_3CO)_2O}$ $\xrightarrow{\text{CH}_3COONa, }\Delta$

(iv) Canizzarro reaction

(A) p-iii, q-ii, r-iv, s-i

(B) p-iii, q-i, r-iv, s-ii

(C) p-ii, q-iii, r-iv, s-i

(D) p-iii, q-ii, r-i, s-iv

				[10						
Q.60		Which one of the following undergoes reaction with 50% sodium hydroxide solution to give the corresponding alcohol and acid?								
	(A)	Phenol	(B)	Benzoic acid						
	(C)	Butanal	(D)	Benzaldehyde						

PART - III MATHEMATICS

- Q.61 If the mid-point of a chord of the ellipse $\frac{x^2}{16} + \frac{y^2}{25} = 1$ is (0, 3), then length of the chord is
 - (A) $\frac{32}{5}$

(B) 16

- (C) $\frac{4}{5}$
- (D) 12
- Q.62 The number of tangents drawn from point (-5, 3) to the hyperbola $\frac{x^2}{25} \frac{y^2}{9} = 1$ are
 - (A) 0

(B) 1

- (C) 2
- (D) 3
- Q.63 If $x^2 + ax + 10 = 0$ and $x^2 + bx 10 = 0$ have a common root, then $a^2 b^2$ is equal to
 - (A) 10

(B) 20

- (C) 30
- (D) 40
- Q.64 A relation R on the set of complex numbers is defined by $z_1Rz_2 \Leftrightarrow \frac{z_1-z_2}{z_1+z_2}$ is real, then R is an
 - (A) equivalence relation

(B) only reflexive

(C) only transitive

- (D) only symmetric
- Q.65 The roots of the equation $\cot x \cos x = 1 \cot x \cos x$ are
 - (A) $n\pi + \frac{\pi}{4}, n \in I$

(B) $2n\pi + \frac{\pi}{4}, n \in I$

(C) $2n\pi \pm \pi \text{ or } n\pi + \frac{\pi}{4}, n \in I$

(D) $n\pi$, $n \in I$

- The degree of the differential equation $x = 1 + \left(\frac{dy}{dx}\right) + \frac{1}{2!} \left(\frac{dy}{dx}\right)^2 + \frac{1}{3!} \left(\frac{dy}{dx}\right)^3 + \dots$ Q.66
 - (A) 3

(B) 2

(C) 1

- (D) 0
- The equation of one of the curves whose slope of tangent at any point is equal to y + 2x is Q.67
 - (A) $y = 2(e^x + x 1)$

(B) $y = 2 (e^x - x - 1)$

(C) $y = 2(e^x - x + 1)$

- (D) $y = 2 (e^x + x + 1)$
- If $I_1 = \int_0^{3\pi} f(\cos^2 x) dx$ and $I_2 = \int_0^{\pi} f(\cos^2 x) dx$ then
 - (A) $I_1 = I_2$
- (B) $3I_1 = I_2$
- (C) $I_1 = 3I_2$ (D) $I_1 = 5I_2$
- Q.69 If $(3\sqrt{3} + 5)^7 = P + F$ where P is an integer and F is a proper fraction, then F. (P + F)
 - (A) 3^7

(B) 2^6

- (C) 3^6
- (D) 2^7
- Q.70 Let $A = \left\{ \begin{pmatrix} a_{11} & a_{12} \\ a_{21} & a_{22} \end{pmatrix} : a_{ij} \in \{0, 1, 2\} \text{ and } a_{11} = a_{22} \right\}$, Then the number of singular matrices in set A is

- Q.71 If $\vec{A} = \hat{i} 3\hat{j} + 4\hat{k}$, $\vec{B} = 6\hat{i} + 4\hat{j} 8\hat{k}$, $\vec{C} = 5\hat{i} + 2\hat{j} + 5\hat{k}$ and a vector \vec{R} satisfies $\vec{R} \times \vec{B} = \vec{C} \times \vec{B}$, $\vec{R} \cdot \vec{A} = 0$ then $\frac{|B|}{|\vec{R} - \vec{C}|}$ is equal to
 - (A) 1

(B) $\frac{1}{2}$

- (C) 2
- (D) 3

(A) [-3, 3]

(B) $[-3, -2] \cup [2, 3]$

(C) $[-2, 2] \cup [3, 4]$

(D) [-2, 4]

Q.73 Let p: 7 is not greater than 4 and q: Paris is in France be two statements. Then \sim (p \vee q) is the statement

- (A) 7 greater than 4 or Paris is not in France
- (B) 7 is not greater than 4 and paris is not in France
- (C) 7 is greater than 4 and Paris is in France
- (D) 7 is greater than 4 and Paris is not in France

Q.74 If the vectors $\vec{a} = 2\hat{i} + \hat{j} + 4\hat{k}$, $\vec{b} = 4\hat{i} - 2\hat{j} + 3\hat{k}$ and $\vec{c} = 2\hat{i} - 3\hat{j} - \lambda \hat{k}$ are coplanar, then the value of λ is equal to

(A) 2

(B) 1

- (C) 3
- (D) -1

Q.75 A is a square matrix such that $A^2 = A$, then $(I - A)^3 + A$ is equal to

(A) A

- (B) I A
- (C) I
- (D) 3A

Q.76 For the equations

$$x + 2y + 3z = 1,$$

$$2x + y + 3z = 2,$$

$$5x + 5y + 9z = 4$$
,

(A) there is only one solution

(B) there exists infinitely many solutions

(C) there is no solution

(D) more than one solutions

- Q.77 Let A = $\{-1, 0, 1, 2\}$, B = $\{4, 2, 0, -2\}$ and f, g: A \rightarrow B be functions defined by $f(x) = x^2 x$ and $g(x) = 2 \left| x \frac{1}{2} \right| 1$. Then
 - (A) f = g

- (B) f = 2g
- (C) g = 2f
- (D) f = 3g
- Q.78. The number of solutions of the system of equations $Re(z^2) = 0$, |z| = 2 is
 - (A) 4

(B) 3

(C) 2

(D) 1

- $Q.79. \quad \left(1 \frac{\cos 61^{o}}{\cos 1^{o}}\right) \left(1 \frac{\cos 62^{o}}{\cos 2^{o}}\right) \left(1 \frac{\cos 63^{o}}{\cos 3^{o}}\right) \left(1 \frac{\cos 119^{o}}{\cos 59^{o}}\right) \text{ is }$
 - (A) -1

(B) 1

- (C) 2
- (D) -2

- Q.80 If $2 \tan^{-1}(\cos x) = \tan^{-1}(2 \csc x)$, then x is equal to
 - (A) $\frac{\pi}{2}$

(B) $\frac{\pi}{3}$

- (C) $\frac{\pi}{4}$
- (D) $\frac{\pi}{6}$
- Q.81 If a, b, c,d and p are different real number such that $(a^2+b^2+c^2)p^2-2$ $(ab+bc+cd)p+(b^2+c^2+d^2)\leq 0$, then a, b, c and d are in
 - (A) A.P.

- (B) G.P.
- (C) H.P.
- (D) A.G.P.
- Q.82 The area of the parallelogram contained by the lines 4y-3x+1=0, 4y-3x-1=0, 3y-4x+1=0 and 3y-4x+2=0 is $\frac{p}{a}$, where p and q are prime, then the value of p+q is
 - (A) $\frac{24}{7}$

(B) $\frac{48}{7}$

- (C) 9
- (D) 3

- For the function $f(x) = \frac{4}{3}x^3 8x^2 + 16x + 5$, x = 2 is a point of Q.83
 - (A) local maxima

(B) local minima

point of inflexion (C)

- (D) decreasing
- Equation of the circle touching the line x + y = 4 at (1, 3) and intersecting the circle $x^2 + y^2 = 4$ orthoconally is Q.84
 - (A) $x^2 + y^2 x + 2y 15 = 0$

(B) $2x^2 + 2y^2 - x - 9y + 8 = 0$

(C) $2x^2 + 2y^2 - x + y - 22 = 0$

- (D) $x^2 + y^2 x y 6 = 0$
- Bag I contains 3 red and 4 black balls while another bag II contains 5 red and 6 black balls. One ball is drawn at Q.85 random from one of the bags and it is found to be black. The probability that it was drawn from bag II is
 - (A) $\frac{7}{43}$

- (B) $\frac{13}{43}$

- For the binomial distribution $(p+q)^n$ whose mean is 20 and variance is 16, pair (n,p) is Q.86
 - (A) $\left(100, \frac{1}{5}\right)$ (B) $\left(100, \frac{2}{5}\right)$ (C) $\left(50, \frac{1}{5}\right)$

- If $N = \frac{1 \cos 4\beta}{\sec^2 2\beta 1} + \frac{1 + \cos 4\beta}{\csc^2 2\beta 1}$ then $\log_{0.25} N$ is equal to
 - (A) -2

- (C) $-\frac{1}{2}$

Space for rough works

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- Q.88 If the coefficient of x in the expanison of $(1-2x^3+3x^5)\left(1+\frac{1}{x}\right)^n$ is 154, then n can be equal to
 - (A) 6

(B) 7

- (C) 8
- (D) 9

- Q.89 If the function f(x) satisfies $\lim_{x\to 1} \frac{f(x)-3}{x^2-1} = \pi$, then $\lim_{x\to 1} f(x)$ is
 - (A) 1

(B) 2

- (C) 3
- (D) π

- Q.90 $\int e^{x} \left(\frac{1 + \sin x}{1 + \cos x} \right) dx = \text{(where C is integration constant)}$
 - (A) $e^x \tan \frac{x}{2} + C$

(B) $e^x \cot \frac{x}{2} + C$

(C) $e^x \sin x + C$

(D) $e^x \cos x + C$