

School Integrated Program

Class – XI

SAMPLE PAPER

INSTRUCTIONS

[Time: 3 Hours]

[Max Marks: 270]

A. General:

- 1. This booklet is your Question Paper containing 90 questions.
- 2. Blank Papers, Clipboards, Log Tables, slide rules, calculators, cellular phones, pagers and electronic gadgets in any form are not allowed to be carried inside the examination hall.
- 3. The answer sheet, a machine-readable optical mark recognition sheet (OMR Sheet), is provided separately.
- 4. DO NOT TAMPER WITH / MULTIPLE THE OMR OR THE BOOKLET.
- 5. Please fill your roll number correctly in the OMR sheet (answer sheet).
- 6. Both Question Paper and OMR Answer Sheet will be submitted after completion of this examination.

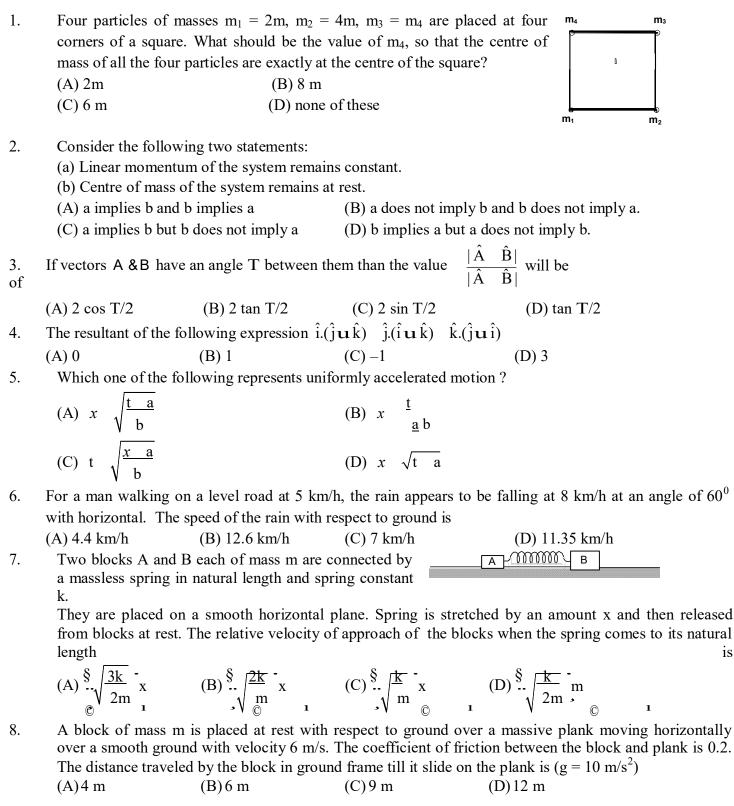
B. Question Paper Format:

- 1. The Question Paper consists of five parts (Part I: MAT, Part II: Physics, Part III: Chemistry, Part IV: Mathematics).
- 2. Each Question carries +3 marks for correct answer and -1 mark for incorrect answer.

Note

- A. Instructions mentioned on this page are of actual test. It has no reference with the questions / pattern of this paper.
- *B.* This paper is provided just to share the pattern, format and level of questions that could be a part of actual test.

PHYSICS



is

CHEMISTRY

9.	Find the oxidation number of S in H_2S_2	20 ₈ H-O-S-O-H	оОн 	
	(A) - 6 $(B) + 6$	(C)+4	(D) none of these	
10.	Correct set of four quantum numbers for the valence (outermost) electron of rubidium			
	(Z = 37) is:			
	(A) 5, 0, 0, +1/2	(B) 5, 1, 0, $+1/2$		
	(C) 5, 1, 1, +1/2	(D)6, 0, 0, +1/2		
11.	The correct order of second ionization	he correct order of second ionization potential of carbon, nitrogen, oxygen and fluorine is		
	(A) C > N > O > F	(B) O > N > F > C		
	(C) O > F > N > C	(D) F > O > N > C		
12.	After balancing the equation in alkaling	alancing the equation in alkaline medium how many OH ⁻ ions would be required?		
	$MnO_4 SnO_2^2 H_2O MnO_2 $ SnO ₃	² OH (alkaline medium	n)	
	(A) 5 (B) 4	(C)3	(D)2	
13.	A toy balloon can occupy 500 ml at 27 [°] C. The minimum stretching capacity of the balloon is three times			
	of this volume at 27 ⁰ C. What is the temperature above which the balloon will burst if pressure of the			
	balloon does not change?			
	(A) 327^{0} C (B) 627^{0} C	$(C) 927^{0}C$	$(D) 1227^{0}C$	
14.	Which of the following is paramagnetic?			
	$(A) O_2$	(B) CN		
	(C) CO	(D) NO		
15.	Among the following, the molecule that is linear is			
	(A) CO_2	(B) NO_2		
	(C) SO_2	(D) ClO_2		
16.	The types of bonds present in CuSO ₄ .5H ₂ O are only			
	(A) electrovalent and covalent			
	(B) electrovalent and coordinate covalent			
	(C) electrovalent, covalent and coordinate covalent			
	(D) covalent and coordinate covalent			
MATHEMATICS				
17.	The least integral value of k such that $(k-2)x^2 + 8x + k + 4$ is positive for all real values of x is			
	(A) 1 (B) 2	(C) 3	(D) 5	
18.	If all the real solutions of the equation	$4^{x} - (a-3)2^{x} + (a-4) =$	=0 are non positive, then	
		4 (C) $a > 4$	(D) a< 3	
19.	The value of 'a' for which the equation	$(a^2 + 4a + 3)x^2 + (a^2 - a - 2a)x^2$	(a + 1)a = 0 has more than two	

19. The value of 'a' for which the equation $(a^2 + 4a + 3)x^2 + (a^2 - a - 2)x + (a + 1)a = 0$ has more than two roots is (A) 1 (P) 2 (C) 2 (D) 1

(A) 1 (B) 2 (C) -2 (D) -1
20. The number of solutions of the equation
$$x^3 + 2x^2 + 5x + 2\cos x = 0$$
 in [0, 2S] is
(A) 0 (B) 1 (C) 2 (D) 3

21. If the equation ax² + bx + c = 0 (a > 0) has two roots D and E such that D < -2 and E > 2, then (A) b² - 4ac > 0 (B) c < 0 (C) a + |b| + c < 0 (D) all of them
22. If log2, log(2^x-1) and log(2^x+3) are in A.P., then the value of x is (A) 5/2 (B) log₂5 (C) log₃5 (D) log₅3
23. If the data given to construct a triangle ABC is a 5, b 7, sin A ³/₄, then it is possible to construct

(B) two triangles(D) no triangle

24.
$$\lim_{\substack{x \\ \bullet 1}} \frac{\sqrt{1 \cos 2(x - 1)}}{(x - 1)}$$

(A) exist and equal to $\sqrt{2}$

(B) exist and equal to - $\sqrt{2}$

(C) does not exist because left hand limit is not equal to right hand limit.

(D) none of these