

Set :- A

Name of the Student: Enrolment No: [11th]

PMT

Time :- 2 Hours

General Instructions

Full Marks :- 225

- 1 This question booklet contains 75 questions. Divided into three sections Section A, Section B and Section C.
- 2 Each section contains 25 multiple choice questions as well as multiple choice question. Choose the most appropriate option.
- 3 Each question carries 3 marks, for each correct answer the student will be awarded 3 marks, zero if not attempted and -1 in all other cases.
- 4 The OMR will be graded by machine so do not fold or make any stray marks on the OMR sheet.
- 5 The bubbles on the OMR sheet should be filled completely with black ball pen. Do not hard press the pen on the OMR sheet.
- 6 Fill the required details in the OMR sheet. Incomplete OMR sheets will not be considered for evaluation.

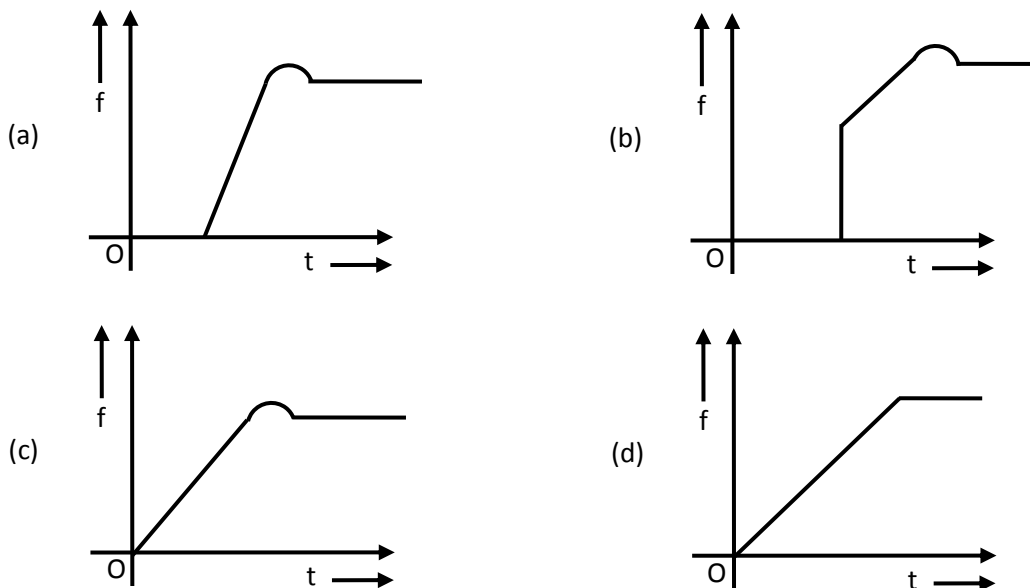
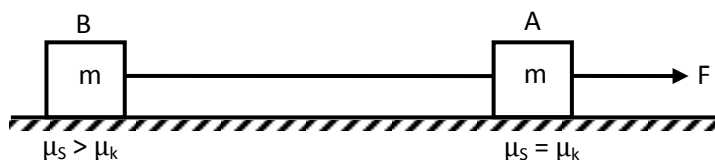
- 1 इस प्रश्न पुस्तिका में 75 प्रश्न शामिल हैं। जो तीन खंडों खंड A, खंड B और खंड C में विभाजित हैं।
- 2 प्रत्येक खंड में 25 प्रश्न शामिल हैं। केवल एक सही विकल्प और एक से अधिक बहुविकल्पीय प्रश्न शामिल हैं। सबसे उपयुक्त विकल्प चुनें।
- 3 प्रत्येक प्रश्न के सही जबाब के लिए 3 अंक मिलेंगे, प्रश्न का हल नहीं करने पर शून्य अंक और गलत विकल्प के लिए -1 अंक मिलेंगे।
- 4 OMR मशीन द्वारा मूल्यांकन किया जाएगा इसलिए OMR शीट पर किसी भी प्रकार का निशान या मोड़ नहीं बनाए।
- 5 OMR शीट पर बने गोले काले बॉल पेन के साथ पूरी तरह से भरा जाना चाहिए। OMR शीट पर कलम से हार्ड प्रेस न करें।
- 6 OMR शीट के दोनो पक्षों में आवश्यक फील्ड भरें। अधूरे OMR शीट का मूल्यांकन नहीं होगा।

Deposit the Question Booklet and OMR sheet both to the invigilator.

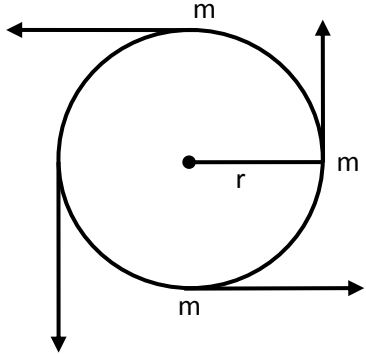
रिजल्ट व अन्य जानकारियाँ OMR शीट में भरे मोबाईल पर SMS से भेजी जाएगी।

SECTION – A

- The displacement of a body is given by $2s = gt^2$ where g is a constant. The velocity of the body at any time t is
 (a) gt (b) $gt/2$ (c) $gt^2/2$ (d) $gt^3/6$
- Two forces 6 N and 3 N are acting on the 2 blocks of 2 Kg and 1 kg kept on frictionless floor. What is the force exerted on 2 kg block by 1 kg block?
 (a) 1 N (b) 2 N (c) 4 N (d) 5 N
- A force $F = t$ is applied to block A as shown in figure. The force applied at $t = 0$ sec when the system was at rest and string is just taut without tension. Which of the following graphs give the friction force between B and horizontal surface as a function of time 't'



- A spring when stretched by 2 mm its potential energy becomes 4 J. it is stretched by 100 mm, its potential energy is equal to
 (a) 4 J (b) 54 J (c) 415 J (d) 100 J

5. In a circus, stuntman rides a motorbike in a circular track of radius R in the vertical plane the minimum speed at highest point of track will be
 (a) $\sqrt{2 Rg}$ (b) $2 Rg$ (c) $\sqrt{3 Rg}$ (d) none of these
6. If the net external force acting on a system is zero, then the centre of mass,
 (a) may accelerate (b) must not accelerate (c) must not move (d) cannot be predicted
7. The spacecraft of mass M moves with a velocity V in free space at first then it explodes breaking into 2 pieces. If after explosion, a piece of mass m comes to rest, the other piece of space craft will have a velocity.
 (a) $\frac{MV}{(M-m)}$ (b) $\frac{MV}{(M+m)}$ (c) $\frac{mV}{(M-m)}$ (d) $\frac{mV}{(M+m)}$
8. A solid iron ball A of radius r collides head on with another stationary solid iron ball B of radius 2r. the ratio of their speeds just after the collision ($e = 0.5$) is
 (a) 3 (b) 4 (c) 2 (d) 1
9. 4 similar particles of mass m are orbiting in a circle of radius r in the same direction and same speed because of their mutual gravitational attractive force speed of particle is given by
 (a) $\left[\frac{Gm}{r} \left(\frac{1+2\sqrt{2}}{r}\right)\right]^{1/2}$
 (b) $\left[\frac{Gm}{r}\right]^{1/3}$
 (c) $\left[\frac{Gm}{r} (1 + 2\sqrt{2})\right]^{1/2}$
 (d) zero
- 
10. A force $\vec{F} = 4\hat{i} - 10\hat{j}$ acts on a body at a point having position vector $-5\hat{i} - 3\hat{j}$ relative to origin. The torque on the body about the origin is
 (a) $38\hat{f}$ (b) $-25\hat{f}$ (c) $62\hat{f}$ (d) none of these

11. A boy sitting firmly over a rotating stool has his arms folded. If he stretches his arms, his angular momentum about the axis of rotation
 (a) increases (b) decreases (c) remains unchanged (d) cannot be predicted
12. A particle moves with a constant velocity parallel to the Y – axis. Its angular momentum about the origin
 (a) is zero (b) goes on increasing (c) goes on decreasing (d) remains constant
13. A solid sphere, a hollow sphere and a solid cylinder, all having equal mass and radius, are placed at the top of an incline and released, the friction coefficients between the object and the incline are equal but not sufficient to allow pure rolling. Greatest time will be taken in reaching the bottom by
 (a) the solid sphere (b) the hollow sphere
 (c) the solid cylinder (d) all will take same time
14. The viscous force is equal to (symbols have their usual meanings)
 (a) $6\pi r\eta v$ (b) $6\pi\eta r v$ (c) $3\pi\eta v$ (d) $6\pi\eta v$
15. A body executing SHM passes through its equilibrium. At this instant, it has
 (a) Maximum potential energy (b) Maximum kinetic energy
 (c) Minimum Kinetic energy (d) Maximum acceleration
16. A particle performing SHM on the y-axis according to eqⁿ = A + B sin ωt . Its amplitude is
 (a) A (b) B (c) A + B (d) $\sqrt{A^2 + B^2}$
17. A pendulum clock that keeps correct time on the earth is taken to the moon. It will run
 (a) at correct rate (b) 6 times faster
 (c) $\sqrt{6}$ times faster (d) $\sqrt{6}$ times slower
18. 2 waves of amplitude A_1 and A_2 respectively and equal frequency travel through same point. The amplitude of the resultant wave is
 (a) $A_1 + A_2$ (b) $A_1 - A_2$ (c) between $A_1 + A_2$ and $(A_1 - A_2)$
 (d) can not say

19. When a sound wave is reflected from a wall the phase difference between the reflected and incident pressure wave is
 (a) 0 (b) π (c) $\pi/2$ (d) $\pi/4$
20. Which of the following does not affect the apparent frequency in Doppler effect?
 (a) speed of source (b) distance between source
 (c) frequency of source (d) distance between source and observer
21. Molar heat capacity at constant pressure, $C_p = ?$
 (a) $C_v - R$ (b) $C_v \times R$ (c) $C_v + R$ (d) none of these
22. In a sample of an ideal gas, the average momentum of a molecule depends on
 (a) pressure (b) mass of gas (c) no. of moles (d) none of these
23. Keeping the no. of moles, volume and pressure the same, which of the following are the same for all ideal gas?
 (a) rms speed of a molecule (b) density
 (c) temperature (d) average of magnitude of momentum
24. Boiling water is changing into steam. Under this condition, the specific heat of water is
 (a) zero (b) one (c) infinite (d) less than one
25. Which law can be understood in terms of Stefan's law?
 (a) Wien's displacement law (b) Kirchhoff's law
 (c) Newton's law of cooling (d) Planck's law

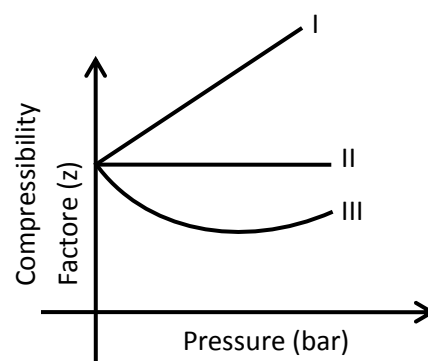
SECTION – B

- 4.4 g of CO₂ and 2.24 lt of H₂ and mixed in a container. The total no. of molecules present in the container
 (a) 6.022×10^{23} (b) 1.2044×10^{23} (c) 6.023×10^{26} (d) 6.023×10^{24}
- Which of the following has greatest no. of atoms?
 (a) 1 g of C₄H₁₀ (b) 1 g of N₂ (c) 1 g of (Ag) (d) 1 g of H₂O
- The no. of nodal plane in P_x orbital is
 (a) 1 (b) 2 (c) 3 (d) 4
- Ionisation energy of He⁺ is 19.6×10^{-18} J/atom. The energy of first stationary state (n = 1) of Li²⁺ is
 (a) -2.2×10^{-15} J/atom (b) 8.82×10^{-17} J/atom (c) 4.41×10^{-16} J/atom (d) -4.41×10^{-17} atom
- The Boyle temperature of three gases are given in the table

Ethene	735 K
Oxygen	400 K
Hydrgoen	110 K

 If the compressibility factor was measured at 400 K, the gases are
 (a) I – ethene, II oxygen, III hydrogen
 (b) I – hydrogen, II ethene, III oxygen
 (c) I hydrogen, II oxygen, III ethene
 (d) I oxygen, II ethene, III hydrogen
- Which of the following exhibits weakest intermolecular forces?
 (a) NH₃ (b) HCl (c) He (d) H₂O
- An example of extensive property is
 (a) Temperature (b) internal energy (c) viscosity (d) molar heat capacity
- For the reaction

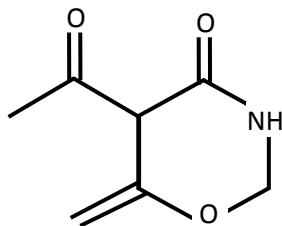
$$\text{H}_2(\text{g}) + \text{I}_2(\text{g}) \rightleftharpoons 2\text{HI}(\text{g}),$$
 The equilibrium constant K_p changes with
 (a) total pressure (b) catalyst
 (c) amount of H₂ & I₂ taken (d) Temperature



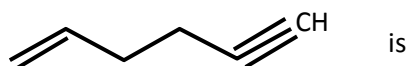
9. In the reaction $\text{AlCl}_3 + \text{Cl}^\ominus \rightarrow \text{AlCl}_4^\ominus$, AlCl_3 . Can be classified as
 (a) acid (b) base (c) a salt (d) none of these
10. Which of the following salt when dissolved in water will hydrolyse
 (a) NaCl (b) KCl (c) NH_4Cl (d) Na_2SO_4
11. Oxidation no. of Mn in KMnO_4
 (a) +4 (b) +7 (c) -4 (d) +3
12. Which one of the following is a reducing agent?
 (a) zero (b) chlorine (c) FeCl_3 (d) Na_2SO_3
13. Oxidation state of phosphorous varies from
 (a) -1 to +1 (b) -3 to +3 (c) -3 to +5 (d) -5 to +1
14. Oxidation number of P in $\text{Mg}_2\text{P}_2\text{O}_7$
 (a) +3 (b) +2 (c) +5 (d) -3
15. No. of lone pair present in oxygen in H_2O is
 (a) 2 (b) 1 (c) 3 (d) None
16. Which of the following is paramagnetic?
 (a) O_2 (b) He (c) N_2 (d) H_2
17. What is the hybridisation of the central atom in NH_3
 (a) sp^2 (b) sp^3 (c) sp (d) sp^3d
18. In the context of carbon, which of the following is arranged in correct order of electronegativity?
 (a) $sp > sp^2 > sp^3$ (b) $sp^3 > sp^2 > sp$ (c) $sp^2 > sp > sp^3$ (d) $sp^3 > sp > sp^2$
19. Which of the following mainly forms superoxide on reaction with oxygen
 (a) Na (b) K (c) Ca (d) N

20. $\text{CaCO}_3 \cdot 2\text{H}_2\text{O}$ is commonly known as
 (a) Plaster of paris (b) Gypsum (c) Epsom salt (d) Dolomite

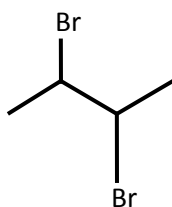
21. Identify which function group is not present in following compound



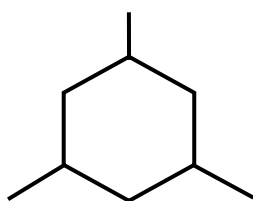
- (a) ketone (b) Ester (c) Amide (d) Ether
22. The IUPAC name of



- (a) Hex – 5 – en – 1 – yne (b) Hex – 1 – en – 5 – yne
 (c) Hex – 6 – en – 1 – yne (d) Hex – 1 – en – 6 – yne
23. How many secondary carbon and hydrogen atoms are present in the molecule given below



- (a) 2, 3 (b) 2, 2 (c) 3, 3 (d) 2, 0
25. Total no. of stereoisomer formed by the given compound



- (a) 2 (b) 3 (c) 4 (d) 8

25. Bond angle between C—H in CH_4

- (a) 90° (b) $109^\circ 29'$ (c) $104^\circ 30'$ (d) $107^\circ 8'$

SECTION – C

1. When a potted plant was cut few inches above soil, then water oozed out of the cut part. It was due to
 (a) transpiration (b) root pressure
 (c) capillary (d) none of above.
2. Path of water movement from soil to xylem is :
 (a) metaxylem→protoxylem→cortex→soil→root hair
 (b) cortex→root hair→endodermis→pericycle→protoxylem→metaxylem
 (c) soil → root hair→cortex→endodermis→pericycle→protoxylem→metaxylem
 (d) pericycle→soil→root hair→cortex→endodermis→protoxylem→metaxylem.
3. Active transport of elements across the cell membrane requires
 (a) ATP (b) acetyl choline (c) phloroglucinol (d) cyclic AMP.
4. Total amount of water present in the soil is called
 (a) chresard (b) holard (c) echard (d) none of the above.
5. In soil ,the water available for root absorption is
 (a) gravitational water (b) capillary water
 (c) hygroscopic water (d) combined water
6. The plasmalemma and the tonoplast is an osmotic system which function as
 (a) semipermeable and selectively permeable membrane
 (b) impermeable membranes
 (c) permeable membranes
 (d) unit membranes
7. wilting of plant occurs when occurs when
 (a) xylem is blocked (b) phloem is blocked
 (c) pith is removed (d) epidermis and few roots are removed.
8. Water absorption in roots mainly takes place in which zone of root?
 (a) zone of elongation (b) root hair zone
 (c) root epidermis (d) maturation zone.

9. Water logged soil is
 - (a) physically dry
 - (b) physiologically dry
 - (c) both physically and physiologically dry
 - (d) nether physically nor physiologically dry.
10. Exudation of xylem sap on cutting of a shoot is due to
 - (a) guttation
 - (b) root pressure
 - (c) transpiration
 - (d) none of the above
11. after heavy rainfall with poor drainage leaves of many plants wilt due to
 - (a) root rot
 - (b) poor soil aeration
 - (c) high salt concentration
 - (d) low sol temperature.
12. Girdling is not possible in oryza sativa because
 - (a) they lack cambium
 - (b) they have scattered vascular bundles
 - (c) they cannot show secondary growth
 - (d) all above
13. Scotoactive stomatal movement can be observed in the leaves of
 - (a) vallisneria
 - (b) potamogeton
 - (c) bryophyllum
 - (d) nerium
14. The most accepted theory for stomatal opening and closing is
 - (a) transpiration
 - (b) guard cell photosynthesis
 - (c) K + efflux and influx
 - (d) starch glucose interconversion
15. Upward movement of water through xylem is best explain by
 - (a) cohesion tension theory
 - (b) pulsation theory
 - (c) capillarity theory
 - (d) root pressure theory
16. Metal associated with an enzyme act as
 - (a) activator
 - (b) terminator
 - (c) translator
 - (d) converter
17. A working combination of an apoenzyme and a coenzyme is termed as
 - (a) prosthetic group
 - (b) holoenzyme
 - (c) enzyme-substrate complex
 - (d) enzyme product complex
18. K_m value of an enzyme is substrate concentration at
 - (a) $1/2 V_m$
 - (b) $1/4 V_m$
 - (c) $3/2 V_m$
 - (d) $4/5 V_m$

19. The metazoans, without tissue grade of organization, are called
 (a) parazoa (b) protozoa (c) eumetazoa (d) deuterostomia
20. The organisms attached to the substratum, generally, possesses
 (a) radial symmetry (b) one single opening of digestive canal
 (c) asymmetrical body (d) cilia on surface to create water current
21. Alimentary canal is not found in
 (a) apoda (b) cestoda (c) gastropoda (d) arachnida
22. Which of the following is generally given as a wedding gift in Japan ?
 (a) Euglena (b) Euspongia (c) Spongilla (d) Euplectella
23. The power of regeneration in sponges is due to
 (a) thesocytes (b) amoebocytes (c) scleroblasts (d) archaeocytes
24. In Hydra, the undigested waste material and nitrogenous waste material is removed from
 (a) mouth and body wall (b) mouth and tentacles
 (c) mouth and nematocyst (d) body wall and tentacles
25. Flatworms excrete through
 (a) kidney (b) nephridia (c) protonephridia (d) Malphigian tubules