

1. The value of the closed surface integral  $\int \mathbf{r} \cdot d\mathbf{S}$  where  $\mathbf{r}$  is the radius vector and  $d\mathbf{S}$  is an element of vector area on a spherical surface of radius  $R$  is
  - A) Zero
  - B) The volume of the sphere
  - C) Total surface area of the sphere
  - D) 3 x volume of the sphere
  
2. The function  $\cot(z)$  has
  - A) An infinite number of poles each having a residue equal to 1
  - B) One pole at  $z = 0$  and the value is 1
  - C) An infinite number of poles each having residue equal to 0
  - D) Exactly one pole at  $\pi/2$  with a residue equal to 1
  
3. A good example of a vector which is solenoidal is
  - A) The magnetic field vector  $\mathbf{B}$
  - B) The electric field vector  $\mathbf{E}$
  - C) The current density vector  $\mathbf{J}$
  - D) The velocity vector  $\mathbf{V}$  of an electron in an atomic orbit
  
4. Consider the step function:  $f(t) = +1$  for  $0 \leq t \leq \pi/\omega$  and  $f(t) = -1$  for  $-\pi/\omega \leq t \leq 0$ . Which of the following series gives the Fourier series representation of this function? (The summations are over  $n = 1$  to  $\infty$ ).
 

A) $(4/\pi) \sum [\sin(n\omega t)/n]$	B) $(4/\pi) \sum [\sin\{(2n+1)\omega t\}/(2n+1)]$
C) $(4/\pi) \sum [\cos(n\omega t)/n]$	D) $(4/\pi) \sum [\cos\{(2n+1)\omega t\}/(2n+1)]$
  
5.  $\mathbf{A}$  and  $\mathbf{B}$  are two arbitrary vectors.  $\epsilon_{ijk}$  is the Levi-Civita symbol in 3 dimensions. Then what is the nature of the quantity  $A_i B_j \epsilon_{ijk}$ ?
 

A) A scalar	B) A tensor of 3 <sup>rd</sup> rank
C) A tensor of 5 <sup>th</sup> rank	D) A vector
  
6. If  $dy/dx = (y/2\sqrt{x})$ , and  $y = 1$  when  $x = 4$ , then
 

A) $y = e^{\sqrt{x}-2}$	B) $\ln y = \sqrt{x-2}$
C) $y = 4\sqrt{x} - 8$	D) $y = e^{\sqrt{x}}$
  
7. Which of the following is not a solution of the Laplace's equation in spherical polar co-ordinates?
 

A) $1/r$	B) $ar^2 + b/r^2$
C) $ar + b/r^2$	D) $b/r^3$
  
8. The Laplace transform of  $e^{-at}t^n$  is
 

A) $n!/(s+a)^{n+1}$	B) $n/(s+a)^n$
C) $(s+a)^n$	D) $as^n$



17. A moon orbits a distant planet in an elliptical orbit. The distance covered by the moon each day
- Is greatest when the moon is nearest to the planet
  - Is greatest when the moon is farthest to the planet
  - Remains the same irrespective of its distance from the planet
  - Remains the same irrespective of its distance from the sun
18. A transformation is effected from  $(q,p)$  to  $(Q,P)$  so that  $Q = aq + bp$  and  $P = cq + dp$ , where  $a,b,c$  and  $d$  are constants. This transformation will be canonical if
- $ad - bc = 0$
  - $ad - bc = 1$
  - $ad + bc = 0$
  - $ad + bc = 1$
19. Two electrons, each with a velocity of  $0.8c$  to an observer, are moving towards each other. The relative velocity of one with respect to the other is :
- $1.6 c$
  - $1.2 c$
  - $0.976 c$
  - $0$
20. A particle in a potential has its Lagrangian given by  $L = \frac{1}{2}m(dr/dt)^2 - k(x^2+y^2)^{1/2}$ . Identify the constant of the motion.
- Z component of the velocity
  - Z component of the linear momentum
  - Z component of the angular momentum
  - Potential energy
21. A particle has a kinetic energy equal to its rest mass energy. Its velocity will be
- $0.5c$
  - $0.866c$
  - $0.75c$
  - $\sqrt{2}c$
22. In one cycle of the radio frequency field a proton in a cyclotron is forced to follow a circular trajectory of radius 1 m by the application of a force of 10 Newtons by applying a suitable magnetic field. The work done on the proton in one cycle is
- 10 J
  - 62.8 J
  - Zero
  - 62.8 W
23. The Poisson bracket for  $[L_x, L_y]$  is
- $iL_z$
  - $L_z$
  - 0
  - $L_x L_y$
24. A hypothetical tunnel with frictionless walls is built connecting two spots on the Earth that follows a straight trajectory through the interior, but does not intersect the center of the Earth. An object is dropped into the tunnel from one end. Assume that the Earth is a uniform spherical mass  $M$  with radius  $R$ , and that the medial point of the tunnel passes to within a distance  $d$  of the center ( $d \ll R$ ). Then
- It will execute simple harmonic motion with a period equal to  $2\pi (\sqrt{R^3/GM})$
  - It will execute simple harmonic motion with a period equal to  $2\pi (\sqrt{d^3/GM})$
  - It will execute simple harmonic motion with a period equal to  $2\pi (\sqrt{dR^2/GM})$
  - It will stop at the centre of the tunnel

25. As a coil is removed from a magnetic field, an emf is induced in the coil which causes a current to flow in it. This current interacts with the magnetic field producing a force which
- Acts at right angles to the direction of motion
  - Acts in the direction of the motion of the coil
  - Acts in the direction opposite to the direction of motion of the coil
  - Causes the coil to flip over
26. A solid metal block is placed in an external electric field. Which statement is correct regarding the electric field inside the block?
- The interior field points at right angles to the direction of the external field
  - The interior field points in the direction opposite to that of the external field
  - The interior field points in a direction parallel to that of the external field
  - There is no electric field inside the block
27. Electric field lines
- Circulate clockwise around positive charges
  - Radiate outward from positive charges
  - Radiate outward from negative charges
  - Circulate counter-clockwise around positive charges
28. Magnetic permeability has units as
- $\text{Wb} / \text{m}^2$
  - $\text{Wb} / \text{A.m}$
  - $\text{A} / \text{m}$
  - $\text{Tesla} / \text{m}$
29. Magnetic susceptibility of diamagnetic materials is of the order of (SI units)
- $+10^{-5}$
  - $-10^{-5}$
  - $+10^5$
  - $-10^{-4}$  to  $+10^{-2}$
30. An electromagnetic wave has an electric field with peak value 250 N/C. What is the average intensity of the wave?
- $0.66 \text{ W/m}^2$
  - $0.89 \text{ W/m}^2$
  - $83 \text{ W/m}^2$
  - $120 \text{ W/m}^2$
31. Which one of the following statements concerning the wavelength of an electromagnetic wave is true?
- The wavelength is independent of the speed of the wave for a fixed frequency
  - The wavelength is inversely proportional to the speed of the wave
  - The wavelength is the same for all types electromagnetic waves
  - The wavelength is inversely proportional to the frequency of the wave
32. Maxwell's equation  $\nabla \cdot \mathbf{B} = 0$  predicts that
- Magnetic monopoles do not exist
  - Magnetic multipoles do not exist
  - Magnetic flux lines are closed curves just like electric flux lines
  - Magnetic flux lines are closed curves unlike electric flux lines



39. An electromagnetic wave passing through a medium of refractive index  $n$  is represented by  $E = 20 \sin [4\pi \times 10^{-4}(x - 2.25 \times 10^9 t)]$ . Here  $x$  is in  $\text{\AA}$  and  $t$  is in nano seconds. What are the values of the wavelength and  $n$ ?
- A)  $5000 \text{ \AA}$  and  $4/3$                       B)  $0.2 \times 10^{-3} \text{ \AA}$  and  $1$   
 C)  $800 \text{ \AA}$  and  $4/3$                       D)  $5000 \text{ \AA}$  and  $3/4$
40. The magnetization of a paramagnetic salt is
- A) Proportional to both the magnetic field and the temperature  
 B) Proportional to the magnetic field and inversely proportional to the temperature  
 C) Proportional to the temperature and inversely proportional to the magnetic field  
 D) Does not depend on the temperature
41. In a purely inductive circuit, the current
- A) Lags behind the emf by  $90^\circ$   
 B) Leads the emf by  $90^\circ$   
 C) Is in phase with the emf  
 D) Sometimes lags behind and sometimes leads the emf
42. The peak in the intensity distribution of a certain emitting surface occurs at  $\lambda = 4000 \text{ \AA}$  at a temperature of  $4000^\circ \text{C}$ . What is the corresponding wavelength at a temperature of  $5000^\circ \text{C}$ ?
- A)  $5000 \text{ \AA}$               B)  $3200 \text{ \AA}$               C)  $3000 \text{ \AA}$               D)  $6250 \text{ \AA}$
43. Sunlight falls on the surface of the earth at the rate of  $1370 \text{ Watts m}^{-2}$ . Assuming the earth to be a perfect absorber of sun's radiations, what will be the pressure exerted by the radiations on the earth?
- A)  $4.63 \text{ Pa}$   
 B)  $4.63 \text{ N.m}^{-2}$   
 C)  $4.63 \times 10^{-6} \text{ N.m}^{-2}$   
 D) Not possible to calculate unless area is specified
44. A statistical energy distribution  $N(E)dE$  gives the number of gas molecules having energies lying in the range from  $E$  to  $E + dE$ . Then the expression for the average kinetic energy over a collection of  $N$  molecules will be
- A)  $(1/N) \int_0^\infty N(E)dE$                       B)  $(1/N) \int_0^\infty EN(E)dE$   
 C)  $(1/N) \int_{-\infty}^\infty EN(E)dE$                       D)  $(1/N) \int_{-\infty}^\infty E dE$
45. Two samples of gas initially at the same temperature and pressure are compressed to half their original volumes, one adiabatically and the other isothermally. The final pressure is
- A) Greater for adiabatic compression  
 B) Greater for isothermal compression  
 C) Same for both  
 D) Half the original pressure for isothermal compression and same as original pressure for the adiabatic compression.

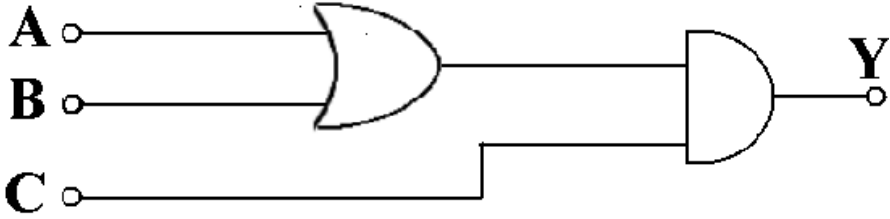
46. Which of the following Maxwell's relations is (are) not true?  
 1.  $(\partial S/\partial V)_T = (\partial P/\partial S)_V$                       2.  $(\partial S/\partial P)_T = -(\partial V/\partial T)_P$   
 3.  $(\partial T/\partial V)_S = -(\partial P/\partial S)_V$                       4.  $(\partial T/\partial P)_S = (\partial V/\partial S)_P$

Choose the correct answer :

- A) 1 only                      B) 2 only                      C) 1 and 2                      D) 3 and 4
47. A famous problem that helped lead to the development of quantum mechanics was the so-called "ultraviolet catastrophe." The "catastrophe" refers to the failure of classical physics to explain what phenomenon?  
 A) The photoelectric effect.  
 B) Electron diffraction.  
 C) Young's double-slit experiment.  
 D) The spectrum of hot, glowing objects.
48. An operator  $U$  is related to the Hamiltonian operator  $H$  through the relation  $U = \exp(i\alpha H)$  where  $H$  is Hermitian and  $\alpha$  is real. Then  $U$  will be  
 A) Real                                      B) Unitary  
 C) Diagonal                                D) Hermitian
49.  $A$  and  $B$  are two non-commuting operators with  $[A, B] = iC$ , then  $C$  will be  
 A) Real                                      B) Unitary  
 C) Diagonal                                D) Hermitian
50. An excited state of the hydrogen atom is specified by the wave function  $A(x+iy)^2 e^{-ar}$ . This will be an eigen state of  $L_z$  having an eigen value of  
 A)  $\hbar$                                       B)  $-\hbar$                                       C)  $2\hbar$                                       D)  $-2\hbar$
51. A particle in an infinite square well is subjected to an additional constant potential  $V_0$ . What will be the first order shift in the energy levels?  
 A)  $V_0$                                       B)  $-V_0$                                       C)  $\frac{1}{2} V_0$                                       D) 0
52. The probability density  $\rho$  and probability current density  $\mathbf{J}$  are related via one of the following relations. Which one is the correct one?  
 A)  $\mathbf{J} = -\nabla\rho$                                       B)  $\nabla \cdot \mathbf{J} + \partial\rho/\partial t = 0$   
 C)  $\mathbf{J} = -D\nabla\rho$                                       D)  $\nabla\rho + \partial\mathbf{J}/\partial t = 0$
53. An electron has the same De Broglie wave length as a proton. Then  
 A) Both have the same linear momentum and have energies in inverse ratio to their masses  
 B) Both have the same linear momentum, but their energies are in the same ratio as their masses  
 C) Both have the same energies, but their linear momenta are in the same ratio as their masses  
 D) Both have the same energies, but their linear momenta are in inverse ratio to their masses

54. The magnetic quantum number of an orbital defines
- The shape of the orbital
  - The energy level of the orbital
  - The spatial orientation of the orbital
  - The spin of the electrons in the orbital
55. In the infinite square well problem, the wave function and its first spatial derivative are:
- Both continuous at the boundaries.
  - Continuous and discontinuous at the boundaries, respectively.
  - Both discontinuous at the boundaries.
  - Discontinuous and continuous at the boundaries, respectively
56. Treating the gravitational interaction between the proton and the electron in a hydrogen atom as a perturbation, the first order correction to the ground state energy of the atom will be
- Of the same order as the energy of the ground state itself.
  - Less than the energy of the ground state by a factor equal to the fine structure constant
  - Proportional to  $\langle 1/r \rangle$
  - Proportional to  $\langle 1/r^2 \rangle$
57. A hydrogen atom is subjected to an electric field of 10 kV/m. The second order Stark shift in the ground state energy of the atom is observed to 0.001 eV. When the electric field is doubled, the shift will change to
- |                     |              |
|---------------------|--------------|
| A) 0.002 eV         | B) 0.004 eV  |
| C) Remains the same | D) 0.0005 eV |
58. The De Broglie wavelength of the electrons in the first Bohr orbit in a hydrogen atom is
- 3.14 Å
  - 1.0 Å
  - Same as the first Bohr radius  $a_0$
  - Equal to the size of the atom
59. In the early days a model for beta decay envisaged the existence of the electron inside the nucleus prior to its emission from the nucleus. Assuming the electron to be confined to within typical nuclear dimensions of 10 fermis and invoking the uncertainty principle, the energy of the emitted beta particle can be estimated to be
- |                            |                 |
|----------------------------|-----------------|
| A) Of the order of 100 MeV | B) Around 1 MeV |
| C) Less than 1 MeV         | D) About 1 keV  |
60. An electron is confined to a thin layer of thickness 1 nm on the surface of a metal by an infinite potential well. It absorbs incident radiation of wavelength 662 nm and jumps from state  $n$  to  $n+1$ . Identify the value of  $n$
- |      |      |      |      |
|------|------|------|------|
| A) 1 | B) 2 | C) 3 | D) 4 |
|------|------|------|------|



61. The energy corresponding to the first excited state of a quantum harmonic oscillator is  
 A)  $\frac{1}{2} \hbar\omega$       B)  $\frac{5}{2} \hbar\omega$       C)  $\hbar\omega$       D)  $\frac{3}{2} \hbar\omega$
62. Which one of the following statements is FALSE ?  
 A) The resistance of intrinsic semiconductor decreases with increase of temperature  
 B) Pure Si doped with trivalent impurities gives a p-type semiconductor  
 C) Majority carriers in a n-type semiconductor are holes  
 D) Minority carries in a p-type semiconductor are electrons
63. To get an output  $Y = 1$  from the circuit shown below, the input must be one of the following. Which one is it?
- 
- A) 1 0 0      B) 0 1 0      C) 0 0 1      D) 1 0 1
64. When a diode is forward biased, the voltage across it  
 A) Is directly proportional to the current  
 B) Is inversely proportional to the current  
 C) Is directly proportional to the source voltage  
 D) Remains approximately the same
65. The center frequency of a band-pass filter is always equal to the  
 A) Bandwidth  
 B)  $-3$  dB frequency  
 C) Bandwidth divided by  $Q$   
 D) Geometric average of the critical frequencies
66. In a transistor the width of the base region is smaller than those of the emitter and collector regions. This is provided  
 A) To ensure that the width of the base region is smaller than the diffusion length of minority carriers in that region  
 B) To minimize the overall dimensions  
 C) So that the emitter and collector are effectively at the same potential  
 D) So that the carriers can tunnel through the base region
67. A bipolar junction transistor is used as a switch. This means that in its operation it switches between  
 A) Cut-off and active region  
 B) Cut-off and saturation region  
 C) Active and saturation region  
 D) Two well separated points in the active region



77. The mid frequency gain of an op-amp amplifier without feedback is 100 and the 3 dB frequencies are 200 Hz and 375 kHz. Negative feedback is introduced with a  $\beta$  of 0.1. The new mid frequency gain and bandwidth are respectively  
 A) 1100 and 34.1 kHz                      B) 9.1 and 4.12 MHz  
 C) 1100 and 4.12 MHz                      D) 9.1 and 34.1 kHz
78. An operational amplifier with a CMRR of 200 gives an output of 5 Volts when the inputs are 99 mV and 101 mV. What will be the output when the inputs are increased to 100 mV and 102 mV respectively?  
 A) Same as 5.0 V since the difference signal is the same.  
 B) 15 V  
 C) 5.01 V  
 D) Cannot calculate unless the difference gain also is given
79. The muon is an elementary particle with the same charge as the electron, but its mass is 206 times larger. Suppose that a muon and a proton come together to form a "muonic hydrogen atom". What will be the ionization energy of muonic hydrogen?  
 A) 2801 eV                      B) 0.95 eV                      C) 0.066 eV                      D) 13.6 eV
80. Given the following term symbols for atoms:  
 1.  $^4S_1$                       2.  $^3P_0$                       3.  $^2P_{3/2}$                       4.  $^2D_{5/2}$   
 Which ones are correct?  
 A) 1 and 2 only                      B) 1 and 4 only                      C) 2, 3 and 4                      D) All
81. An ESR spectrometer operates at 20 GHz frequency. What magnetic field should be used for its correct operation?  
 A) 0.713 T                      B) 7.13 T                      C) 0.0713 T                      D) 713 G
82. Homonuclear diatomic molecules give rise to  
 A) No microwave or IR spectra, but only Raman spectra  
 B) No Raman spectra but microwave or IR spectra  
 C) Both Raman spectra and microwave or IR spectra  
 D) Neither Raman spectra nor microwave or IR spectra
83. In the anomalous Zeeman effect in sodium atom,  
 A)  $D_1$  line and  $D_2$  line split into four lines each  
 B)  $D_1$  line splits into six lines and  $D_2$  line splits into four lines  
 C)  $D_1$  line splits into four lines and  $D_2$  line splits into six lines  
 D)  $D_1$  line splits into four lines and  $D_2$  line remains unaltered
84. In the Raman effect of a given sample, the exciting line has a wavelength of 453.8 nm and a Stokes line is observed at 445.8 nm. At what wavelength does one expect to see the corresponding Anti Stokes line?  
 A) 461.8 nm                      B) 462.1 nm                      C) 437.8 nm                      D) 8 nm

85. The ratio of the frequencies of the light emitted during the  $n = 3$  to  $n = 1$  transitions in a  $\text{Be}^{3+}$  ion and a hydrogen atom is :  
 A) 3 : 1                      B) 4 : 1                      C) 16 : 1                      D) 9 : 1
86. In a gas 10% of the atoms are in the first excited state at a given temperature. Raman scattering is observed from the gas. What will be the ratio of the intensities of the Stokes and Anti Stokes lines?  
 A) 1 : 1                      B) 1 : 9                      C) 3 : 1                      D) 9 : 1
87.  ${}_{26}\text{Fe}^{57}$  has an excited level at 14.4 keV. In a given environment inside a lattice, this level is shifted by  $10^{-7}$  eV. What is the velocity of the Mossbauer drive required to bring this into resonance?  
 A) 2.08 mm/s    B) 2.08 cm/s    C) 4.8 m/s    D) 4.8 mm/s
88. If a hydrogen atom undergoes a transition from the  $n=4$  to the  $n=3$  state, it will:  
 A) Emit a 0.85 eV photon    B) Emit a 0.66 eV photon  
 C) Absorb 0.66 eV of energy    D) Emit a 13.6 eV photon
- 89.. An excited state having a half life of 1 ns decays to the ground state emitting a photon of wavelength  $6000 \text{ \AA}$  in the process. The width of the spectral line will be  
 A)  $0.012 \text{ \AA}$                       B)  $0.361 \text{ \AA}$   
 C)  $0.00361 \text{ \AA}$                       D)  $3.61 \text{ \AA}$
90. The  $K_{\alpha}$  X ray line from an unknown atom is seen to have the same wavelength as the shortest wavelength component of the bremsstrahlung emitted when 57.523 keV electrons bombard a metallic target. Which element is it?  
 A) Helium ( $Z = 2$ )                      B) Lithium ( $Z = 3$ )  
 C) Titanium ( $Z = 22$ )                      D) Tantalum ( $Z = 73$ )
91. The free electrons of a metal  
 A) Do not collide with each other  
 B) Are free to escape through the surface  
 C) Are free to fall into the nuclei  
 D) Are free to move anywhere inside the metal
92. Debye temperature is  
 A) The temperature at which the highest frequency modes (and hence all modes) are excited in a solid.  
 B) A cut off temperature for the vibrations in the solid  
 C) The same as Einstein temperature  
 D) A threshold temperature above which only vibrations can exist inside a solid
93. The energy of the photo electron depends upon the following factor  
 A) Intensity of incident radiation  
 B) Quality of the photocathode  
 C) Frequency of incident radiation  
 D) Type of the incident light source



102. Parallelohedron is a crystal class
- Belonging to the Triclinic system referred to three unequal axes and has a 1-fold symmetry axis
  - Belonging to the Triclinic system referred to three mutually perpendicular axes and has a 2-fold symmetry axis
  - Belonging to the Monoclinic system referred to three mutually perpendicular axes and has a 2-fold symmetry axis
  - Belonging to the Monoclinic system referred to three unequal axes and has a 1-fold symmetry axis
103. Bloch's theorem asserts that the wave function of electrons in periodic potential
- Is an exponentially falling function
  - Is an exponentially increasing function
  - Is in the form of pure plane waves
  - Has the form of a plane wave multiplied by a function with the periodicity of the potential
104. GaAs has a band gap energy of 1.43 eV. This means that the maximum wavelength of the photons that can be absorbed by this semiconductor is
- 867 Å
  - 867 nm
  - 1.43 nm
  - 1430 nm
105. Phonons in aluminium have an energy of 1 eV. The velocity of sound in aluminium is  $6400 \text{ ms}^{-1}$ . What will be the momentum carried by these phonons ?
- $0.04678 \text{ eV} / c$  and is smaller than the momentum carried by photons of same energy
  - $0.04678 \text{ MeV} / c$  and is much larger than the momentum of photons having same energy
  - $2.5 \times 10^{-4} \text{ eV} / c$  and is smaller than the momentum carried by photons of same energy
  - $1 \text{ eV} / c$  and is equal to the momentum carried by photons of same energy
106. The spin and parity of the ground state of  ${}^7_3\text{Li}$  nucleus is
- $(3/2)^-$
  - $(1/2)^+$
  - $0^-$
  - $1^+$
107. Which of the following elementary particles is responsible for carrying away the missing energy and angular momentum in a nuclear beta decay process?
- Electron
  - Electron like neutrino
  - Muon like neutrino
  - Neutron
108. One major difference between alpha decay and beta decay is that
- The alpha particles emitted in alpha decay have continuous energies while the electrons emitted in beta decay have discrete energies.
  - The electrons emitted in beta decay can have continuous or discrete energies while the alpha particles emitted in alpha decay always have discrete energies.
  - The electrons emitted in beta decay have continuous energies while the alpha particles emitted in alpha decay have discrete energies.
  - The electrons emitted in beta decay always have continuous energies while the alpha particles emitted in alpha decay can have continuous or discrete energies.



116. In the compound nucleus theory of nuclear reactions, which of the following statements is not true?
- A) The life time of the compound nucleus is smaller than the time it takes the projectile to pass through the nucleus.
  - B) The compound nucleus can decay in various ways.
  - C) The compound nucleus does not remember the way in which it was formed
  - D) There is complete equilibration of energy among all nucleons.
117. Within the nucleus, the charge distribution
- A) Is constant, but falls to zero sharply at the nuclear radius
  - B) Increases linearly from the centre, but falls off exponentially at the surface
  - C) Increases linearly from the centre, but falls to zero sharply at the nuclear radius
  - D) Is constant, but falls off exponentially at the surface
118. Given the following nuclear masses :  ${}^1_1\text{H}^1$  : 938.783 MeV ;  ${}^4_2\text{He}^4$  : 3728.401 MeV;  ${}^{27}_{13}\text{Al}^{27}$  : 25133.142 and  ${}^{30}_{14}\text{Si}^{30}$  : 27920.387 MeV. Consider the reaction  ${}^1_1\text{H}^1 + {}^{30}_{14}\text{Si}^{30} \rightarrow {}^4_2\text{He}^4 + {}^{27}_{13}\text{Al}^{27}$ . What is the minimum lab energy of protons that can cause the reaction to occur?
- A) 0 MeV
  - B) 2.373 MeV
  - C) The Coulomb barrier energy
  - D) 2.452 MeV
119. An elementary particle whose energy is independent of its velocity is the
- A) Electron
  - B) Proton
  - C) Photon
  - D) Pion
120. Thermonuclear reactions involve reactions between nuclei at energies
- A) Much less than the Coulomb barrier
  - B) In the relativistic range
  - C) Near the threshold energies
  - D) Close to thermal energies of atoms in solids

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